



Neglected Polymorphs of Silica Clouds in Substellar Atmospheres

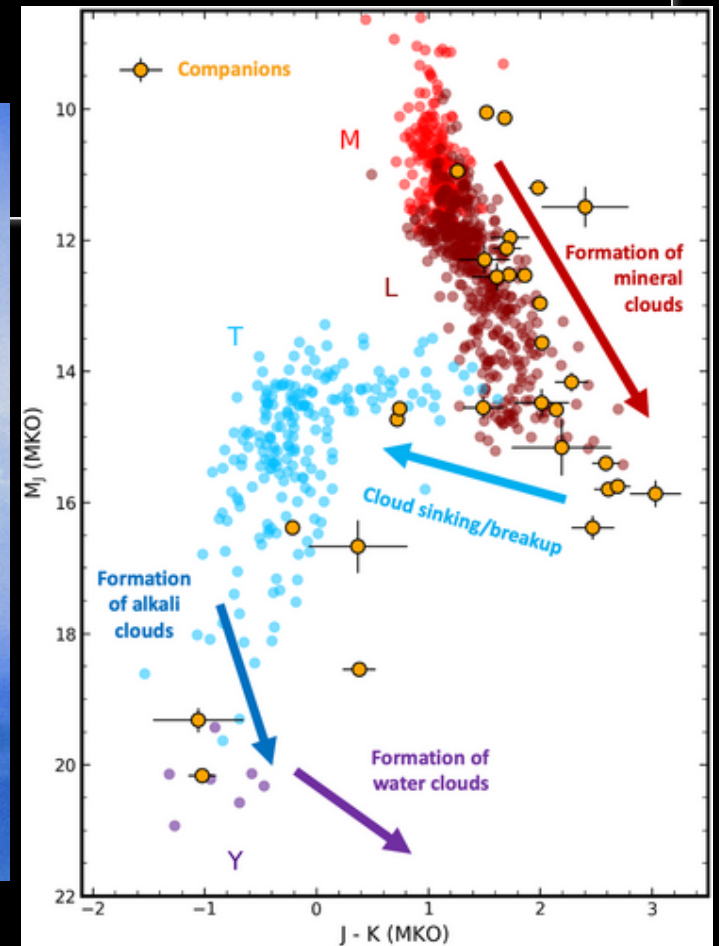
Sarah E. Moran

NHFP Sagan Fellow, NASA Goddard Space Flight Center

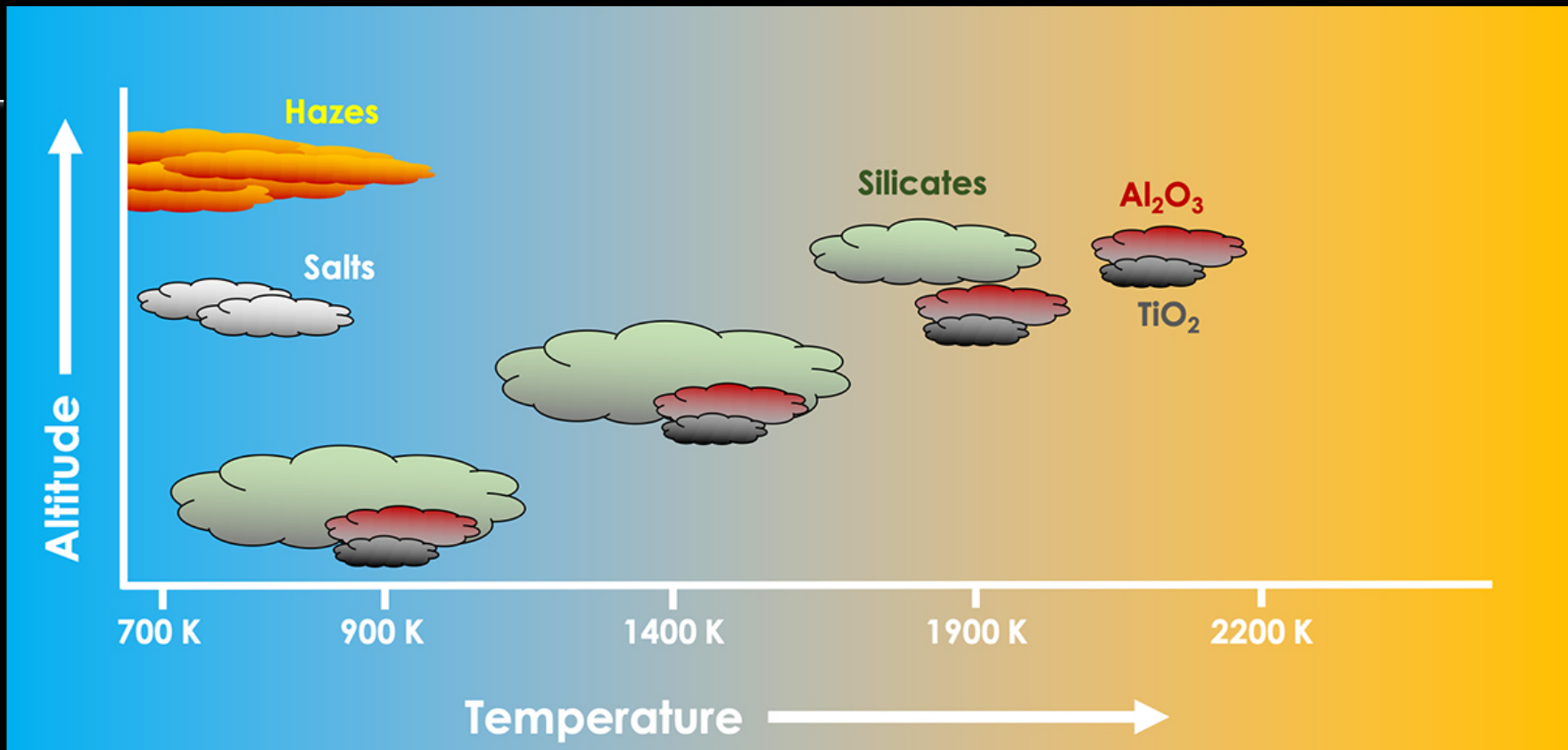
NHFP Symposium || 20 September 2024



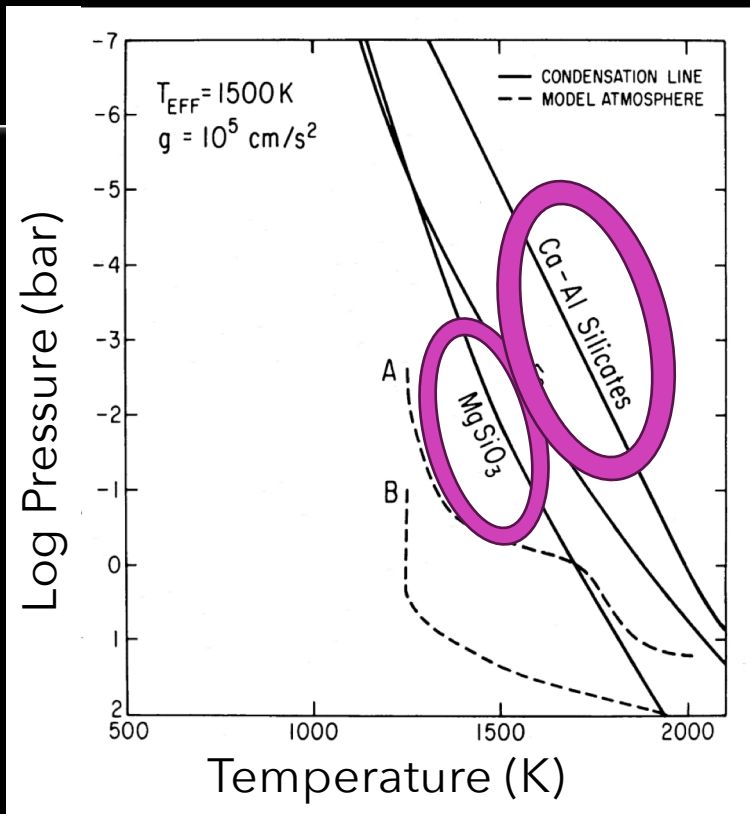
Clouds are a *fundamental* part of brown dwarf and planetary atmospheres



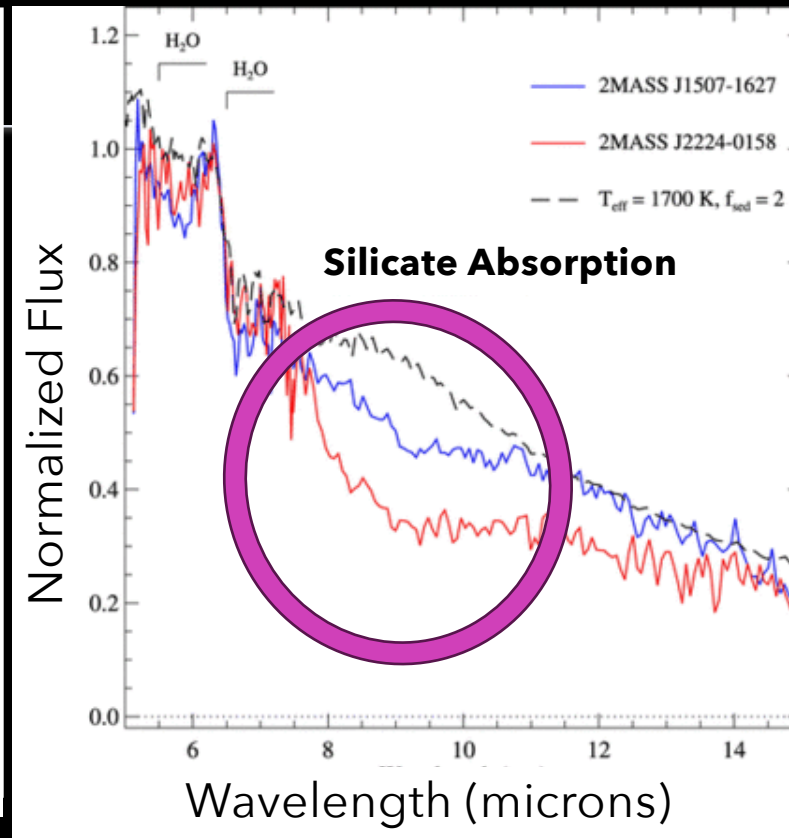
Clouds (and hazes) are **tracers** of the physics and chemistry of these atmospheres



Spitzer **observationally confirmed** that brown dwarf clouds were made of silicates, long **predicted** by theory



Lunine, Hubbard, and Marley, *ApJ*, 1986

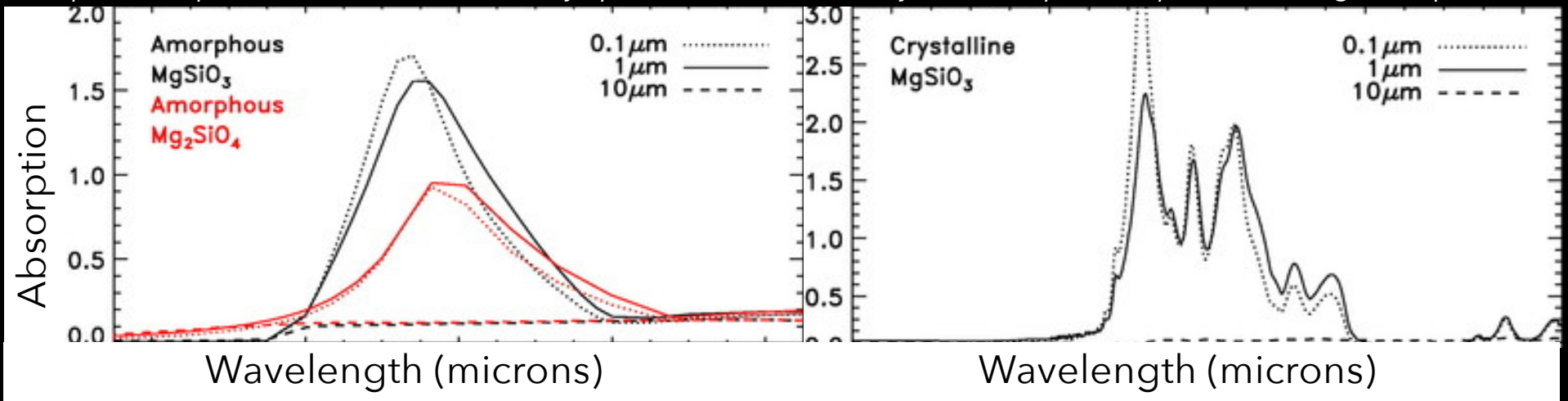


Burgasser et al., *ApJ*, 2008

Amorphous or crystalline particle structure reveals how these clouds formed

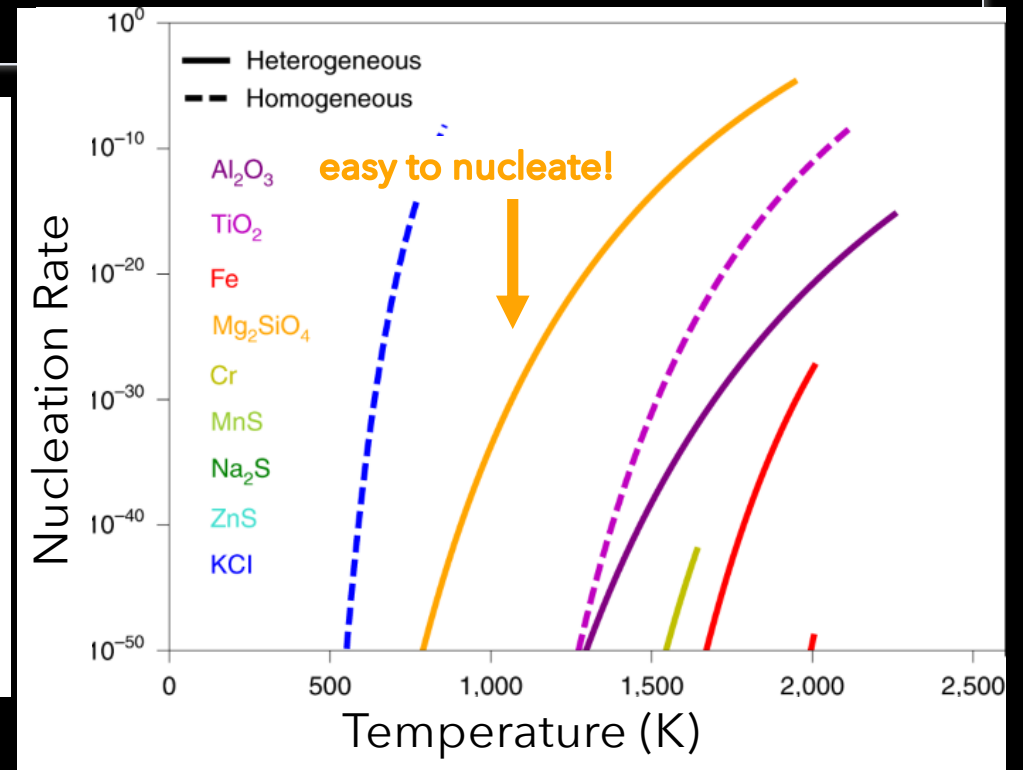
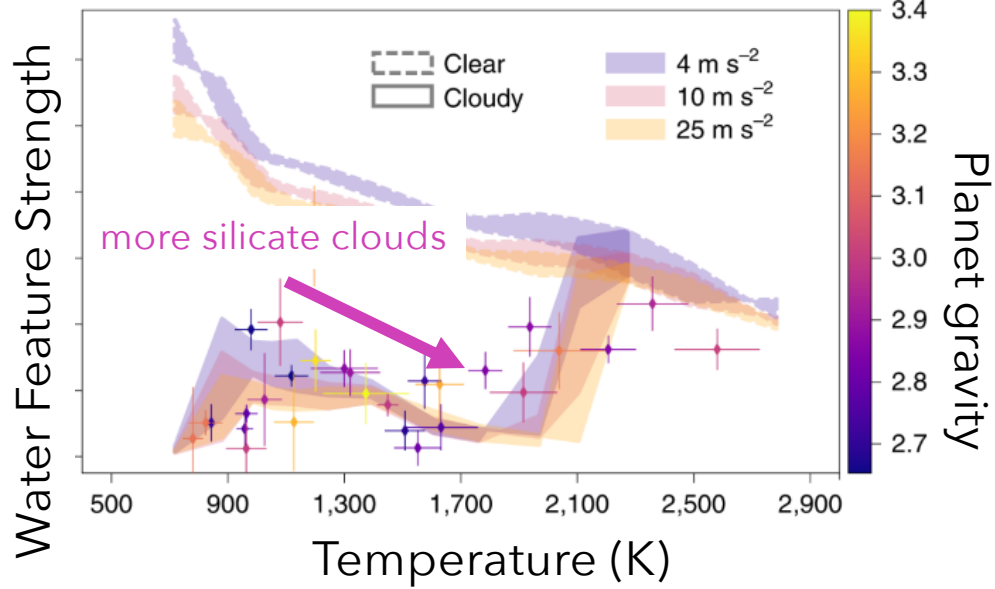
Amorphous = particles form, immediately quenched

Crystalline = particle persists at high temperature

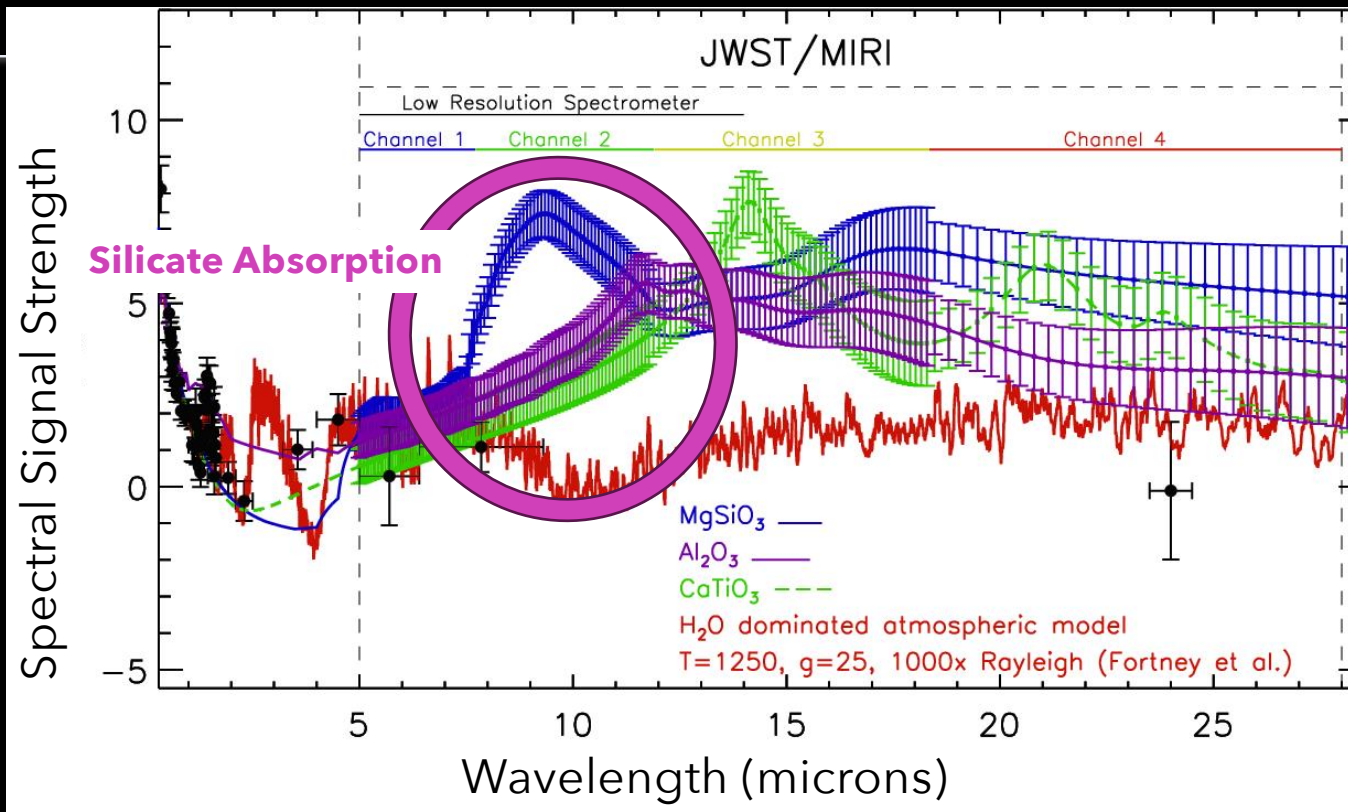


Cushing et al., *ApJ*, 2006

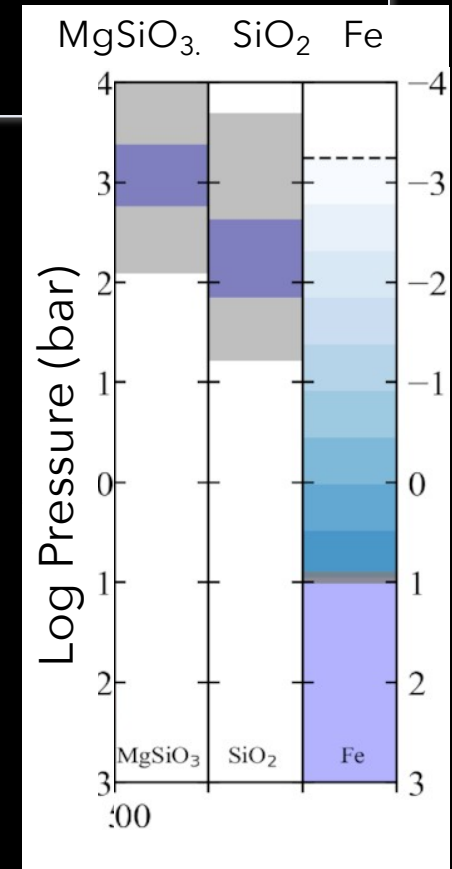
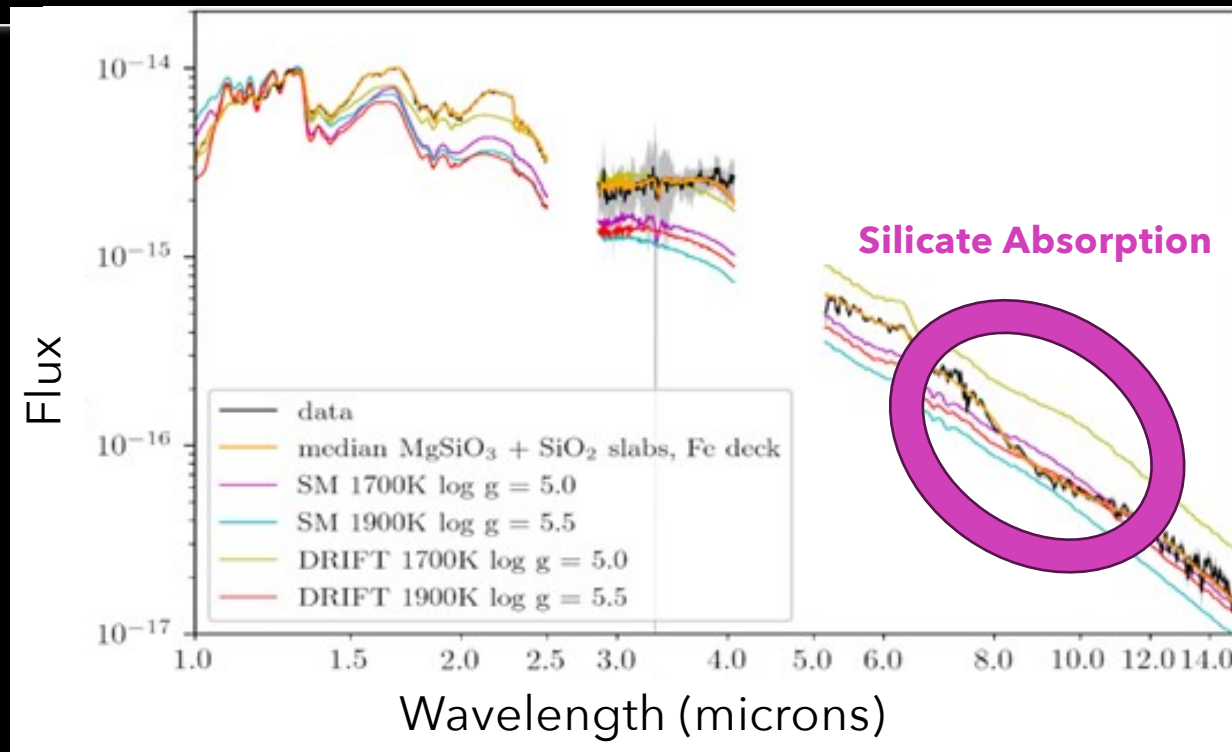
Surveys of hot Jupiters also suggest the dominance of silicate clouds



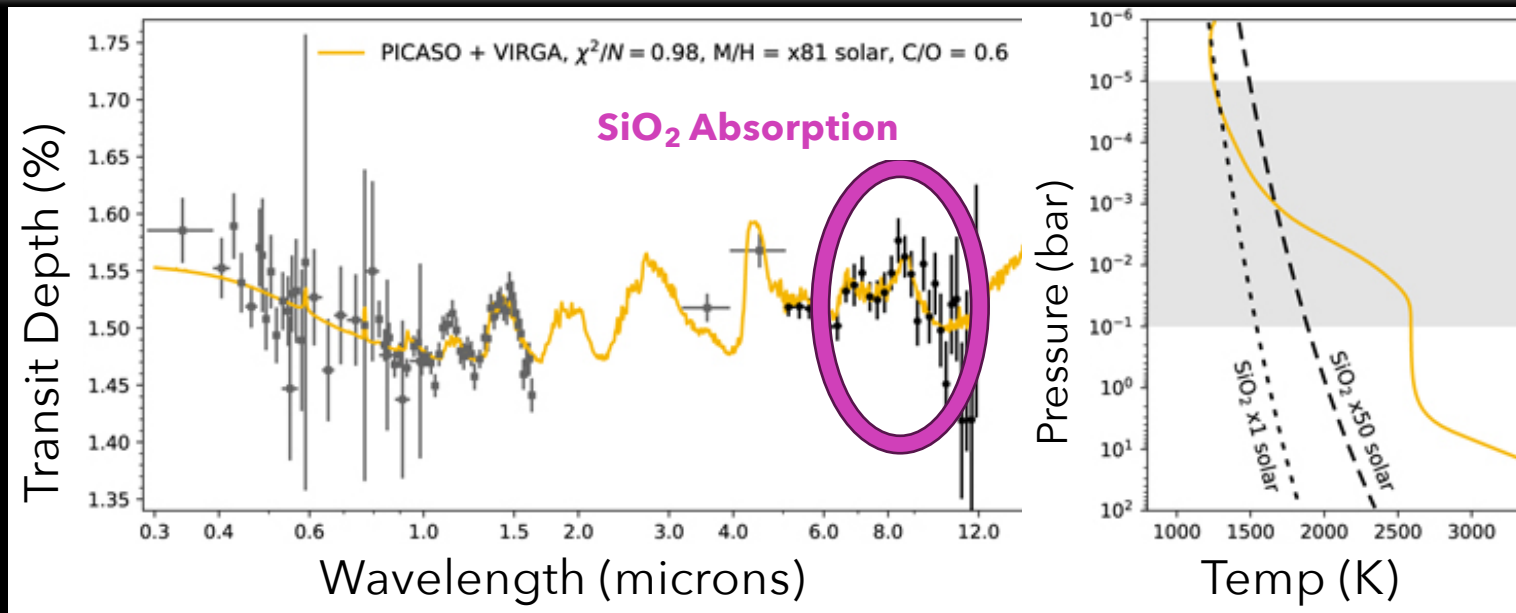
We had lost cooled Spitzer to confirm cloud identities...we needed JWST!



Amorphous SiO_2 , not just MgSiO_3 or Mg_2SiO_4 , is discovered in an L dwarf



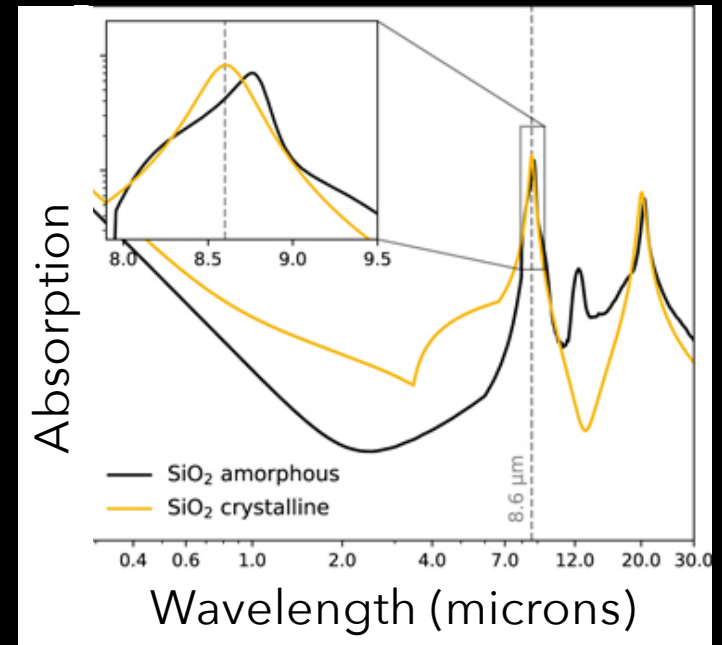
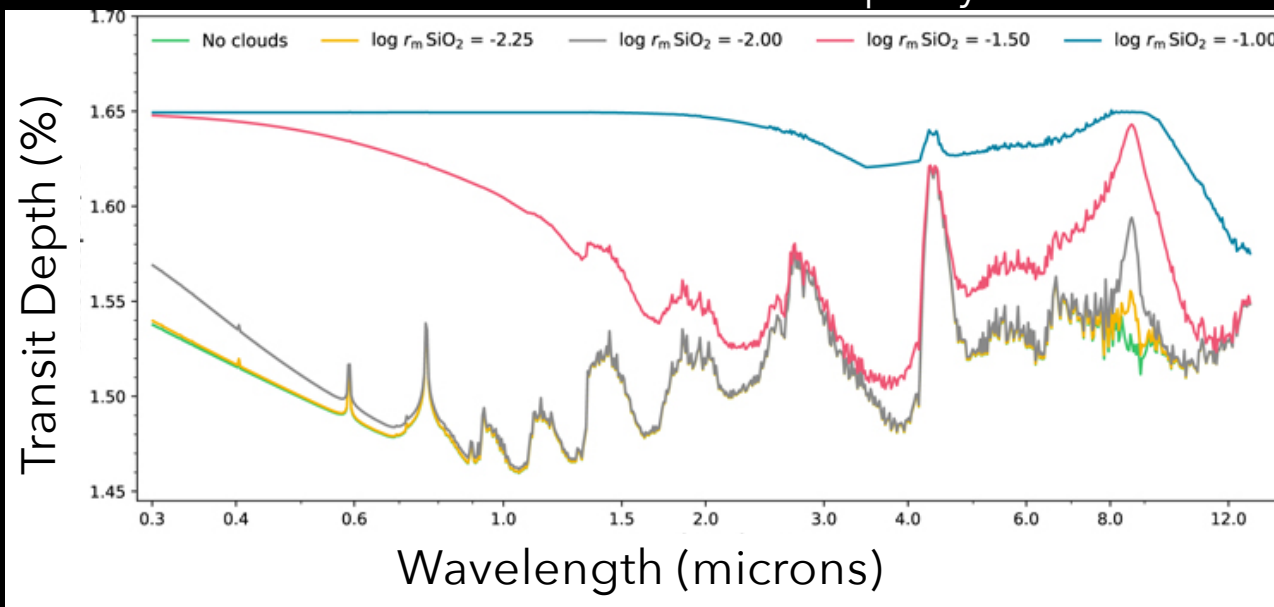
With JWST MIRI LRS, SiO₂ is discovered in the hot Jupiter WASP-17b



Grant et al., *ApJL* 2023

WASP-17b's SiO₂ clouds seem to be **crystalline**, not amorphous, with sub-micron sized particles

Particle size controls the opacity!



Cloud **composition** and **structure** are powerful atmospheric **tracers** of chemistry and physics

oxygen sequestration



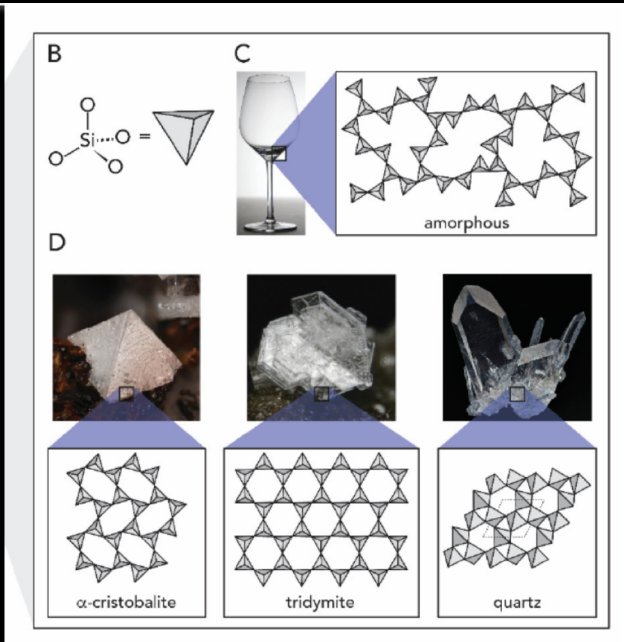
particle size, shape,
and distribution

atmospheric temperature, mixing

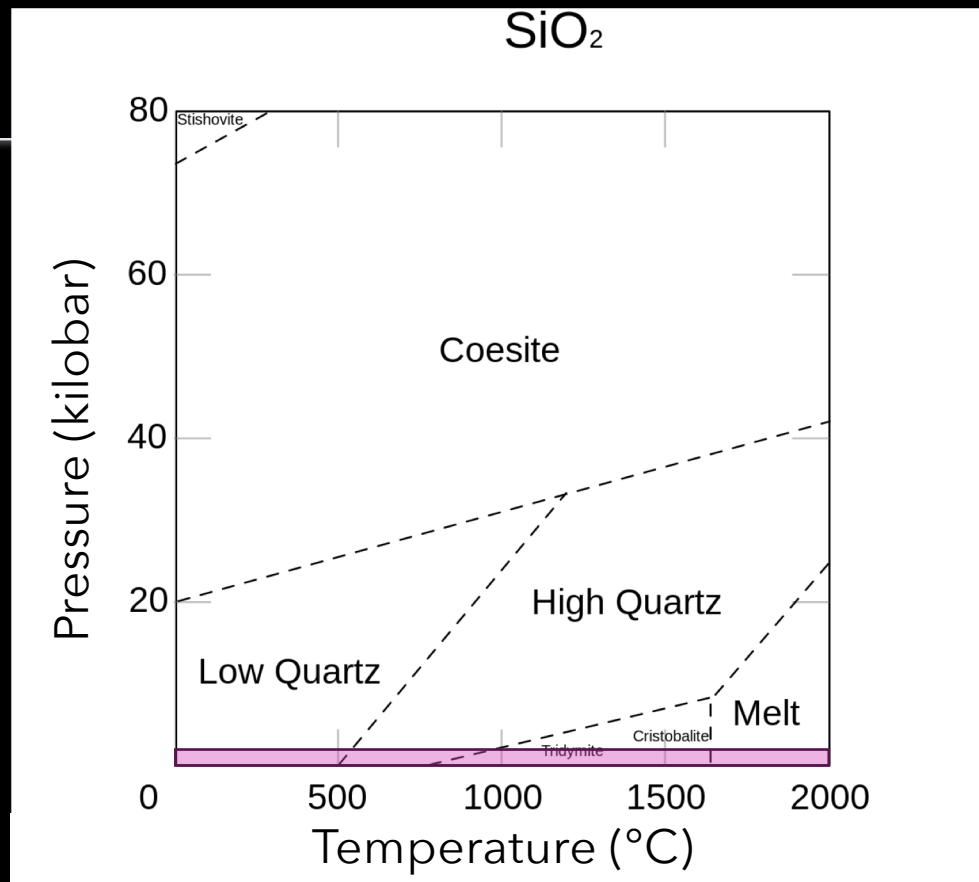
bulk composition of planet and host star

~But we can go deeper~

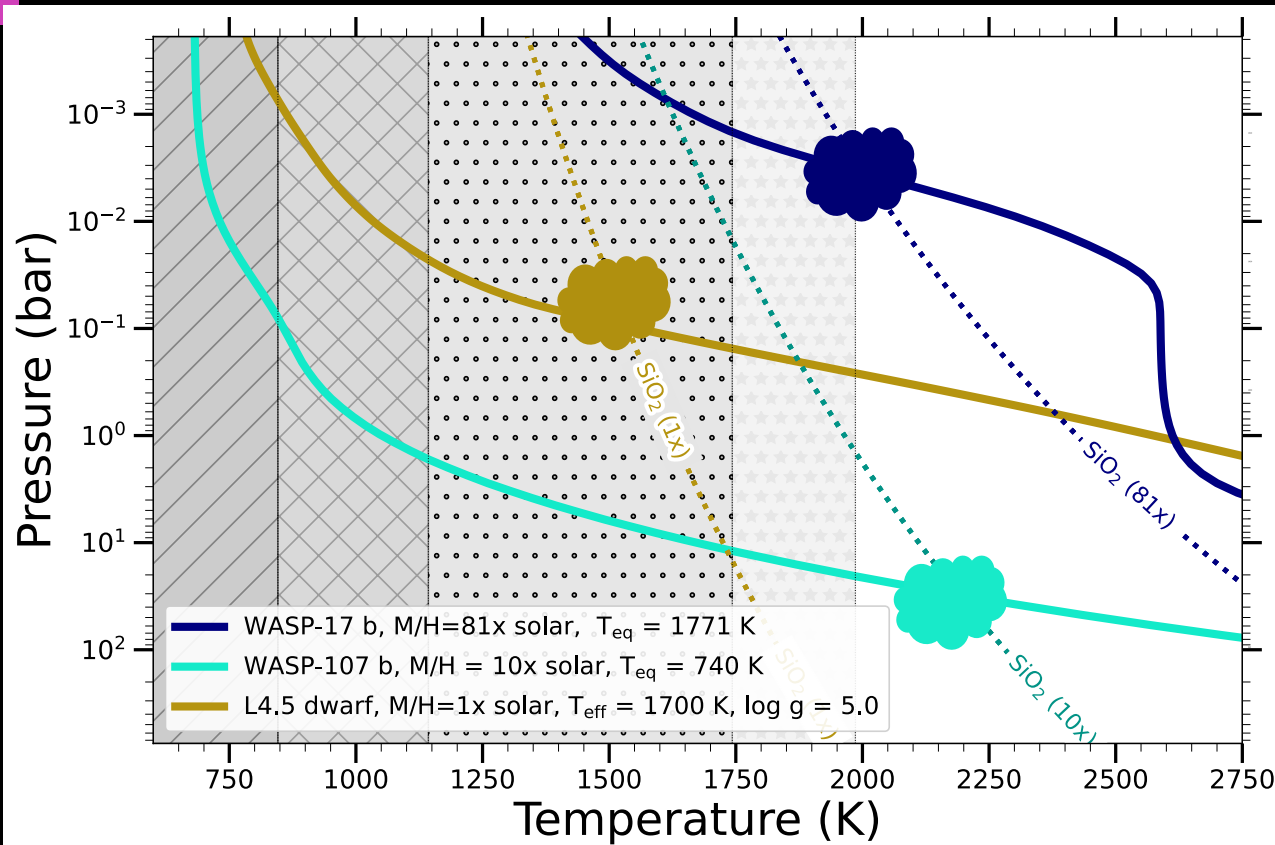
There are multiple stable **polymorphs** of SiO_2 at high temperature/low pressure conditions



Moura et al., 2020, *Biomimetics*

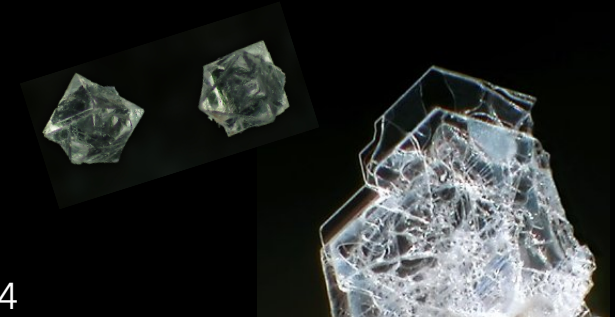


There are multiple stable **polymorphs** of SiO_2 at high temperature/low pressure conditions

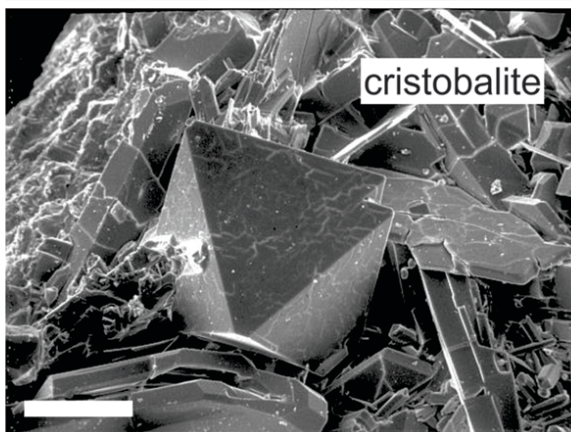
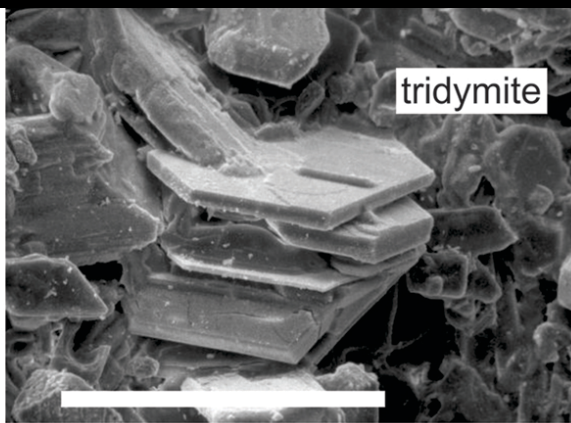
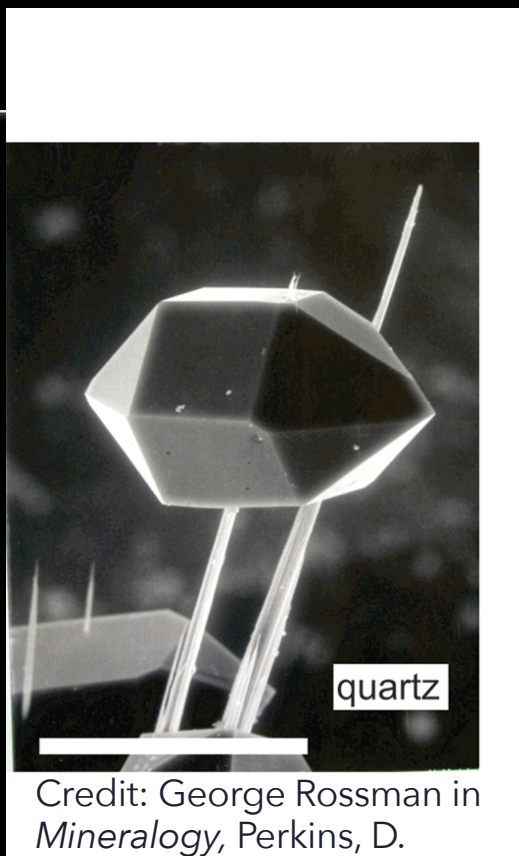


SiO_2 Polymorphs

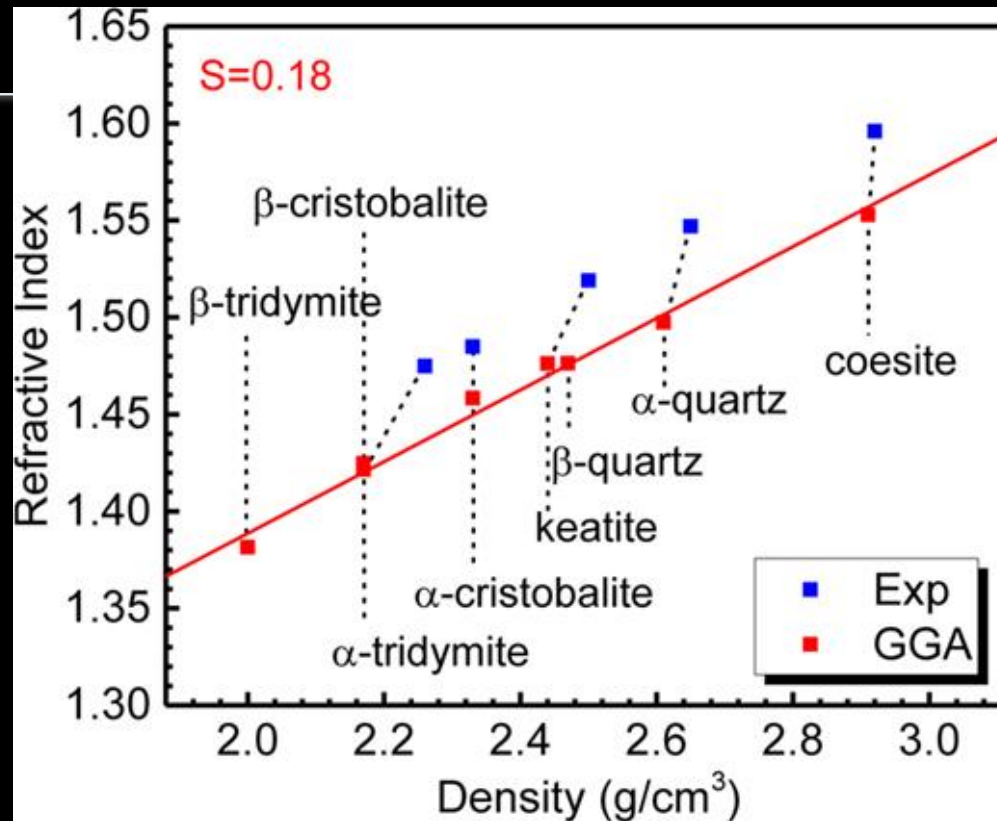
	α -quartz
	β -quartz
	tridymite
	cristobalite



Different crystal structure means different bond lengths, arrangements, and energies



These polymorphs of SiO_2 have different physical and optical properties



Chen et al., *J. Appl. Phys.*, 2023

Do cristobalite and tridymite cloud particles influence substellar atmospheric spectra?

Cloud Model



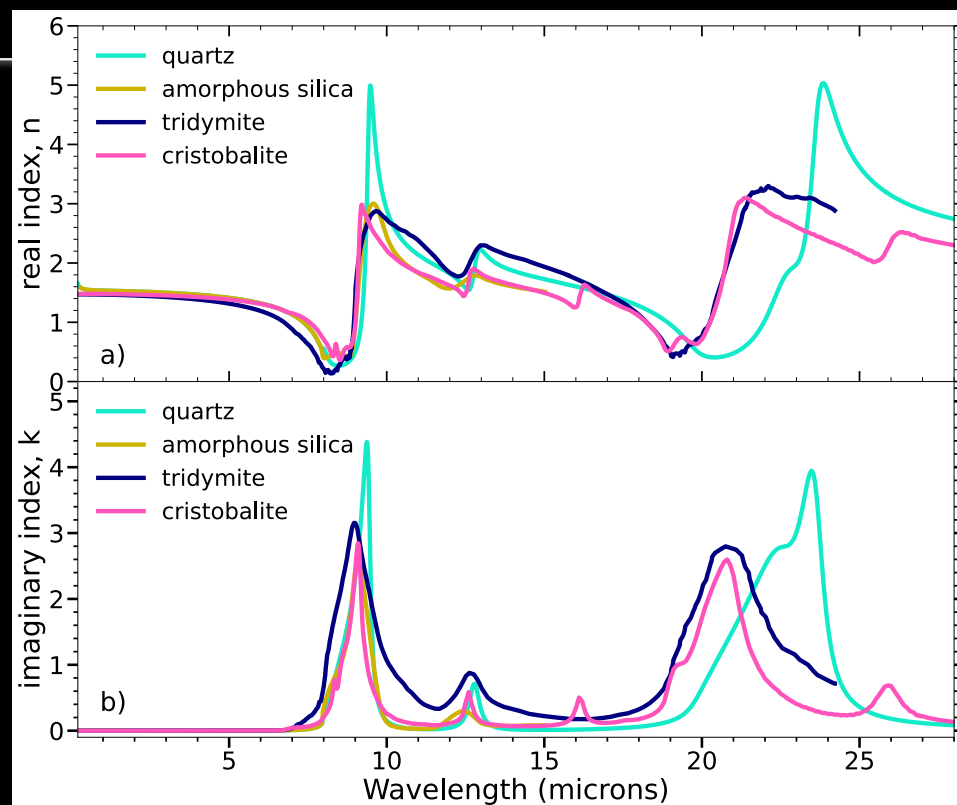
Batalha et al., 2020; Rooney et al., 2022
based on Ackerman and Marley, 2001

Climate, chemistry, radiative transfer



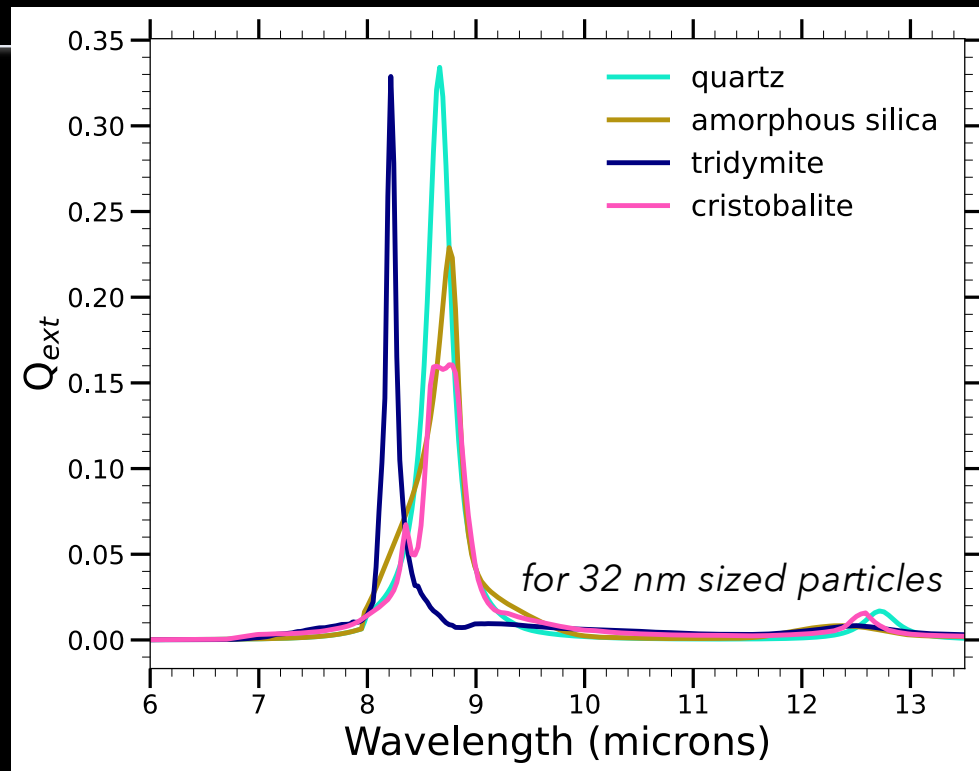
Batalha et al., 2019; Mukherjee et al., 2023
based on Marley et al., 1996

With the available data**, we compile a set of optical properties for tridymite and cristobalite

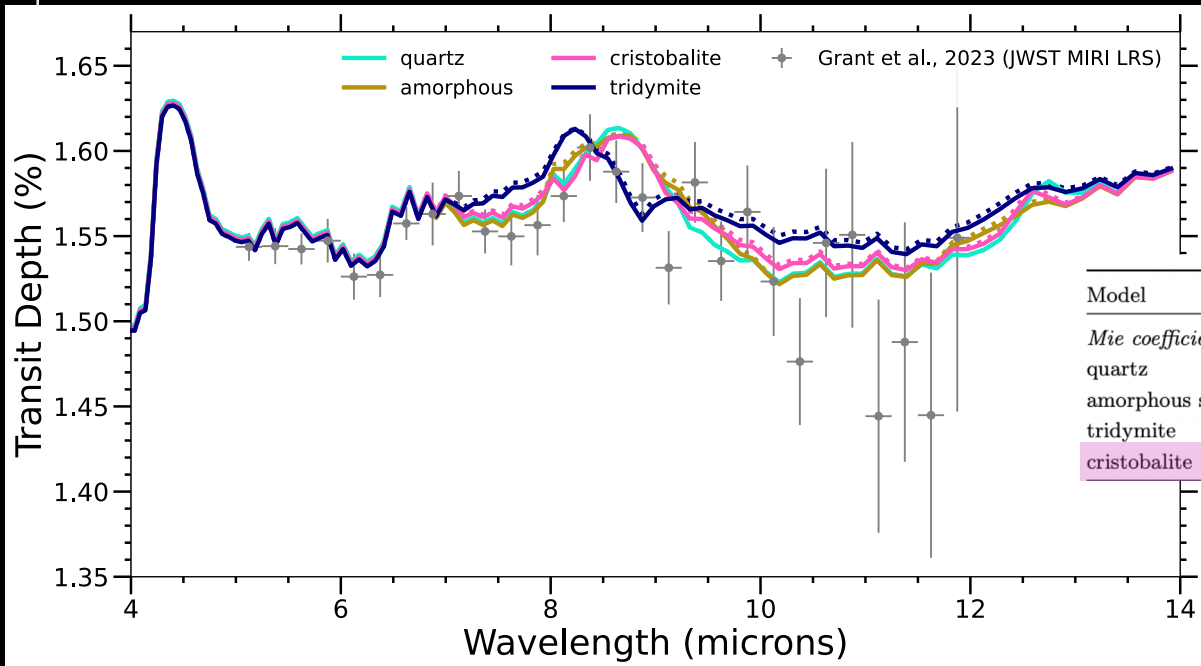


** ask me to expand on this if you want a rant on the state of mineral laboratory data

Mie coefficients are noticeably different at JWST MIRI wavelengths for the various SiO₂ polymorphs



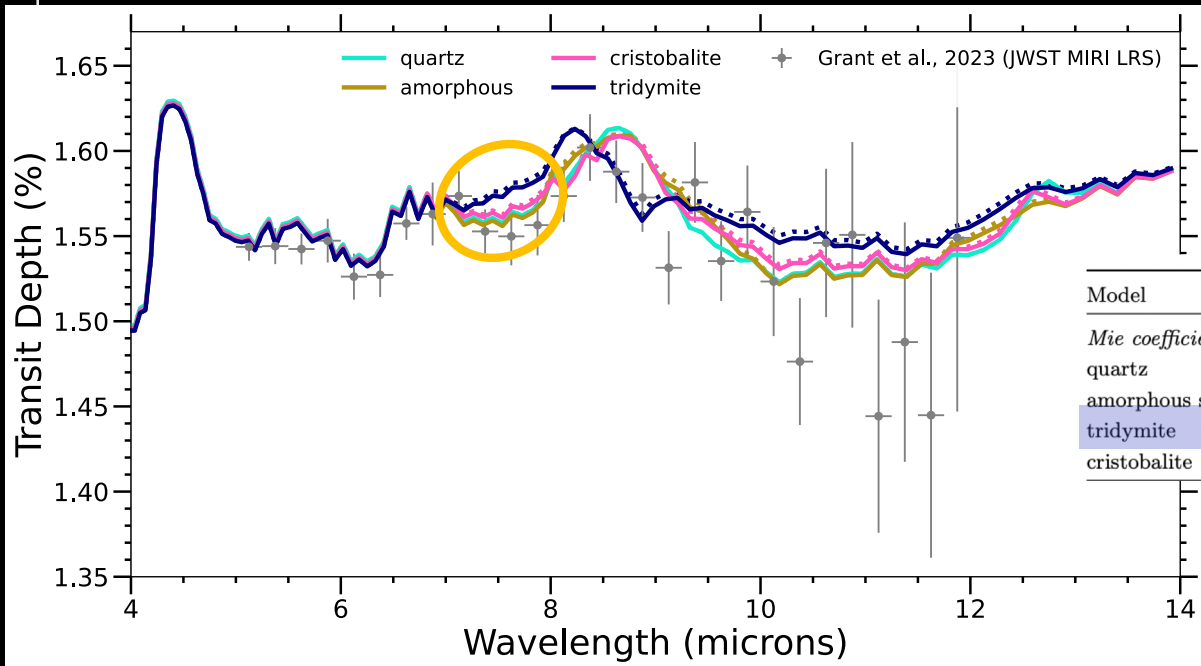
Cristobalite clouds **are a better fit** to the WASP-17 b JWST data compared to either quartz or amorphous silica



Model	χ^2	χ^2/n	BIC	Δ BIC
<i>Mie coefficients, BIC k = 1, n = 95, dof = 94</i>				
quartz	126.3	1.330	130.9	0.6
amorphous silica	126.0	1.327	130.6	0.3
tridymite	131.0	1.379**	135.6	5.3
cristobalite	125.7	1.323	130.3	-

Moran, Marley, Crossley, *ApJL*, 2024

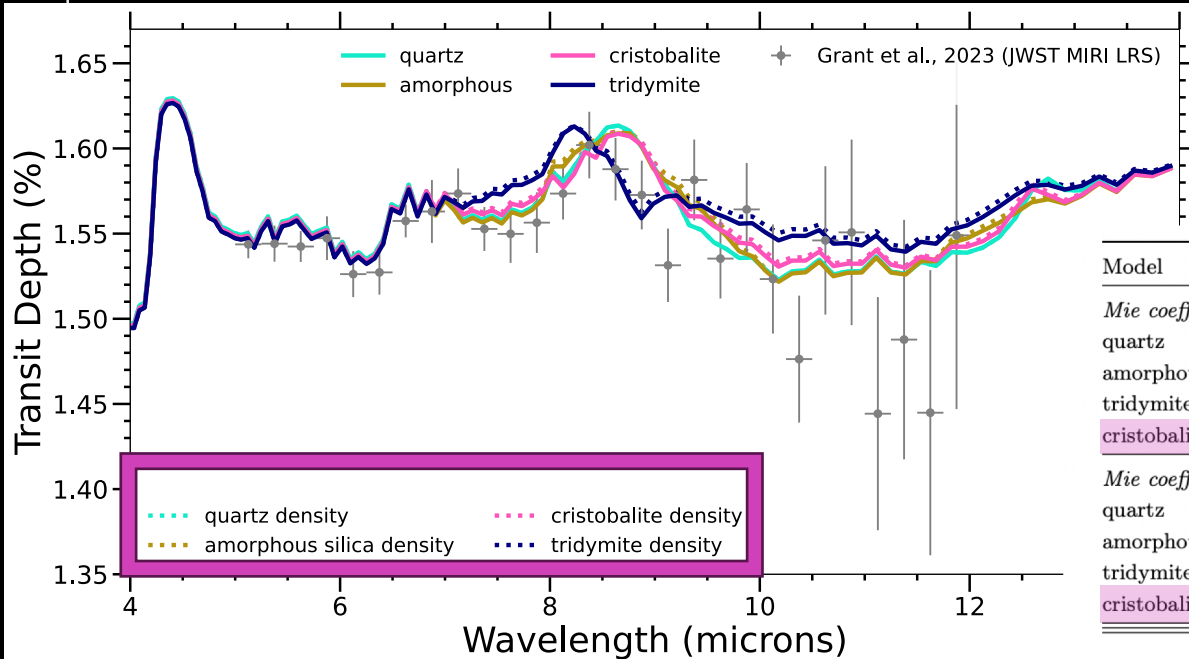
Tridymite clouds **are not a better fit**** to the WASP-17 b JWST data compared to either quartz or amorphous silica



Model	χ^2	χ^2/n	BIC	Δ BIC
<i>Mie coefficients, BIC k = 1, n = 95, dof = 94</i>				
quartz	126.3	1.330	130.9	0.6
amorphous silica	126.0	1.327	130.6	0.3
tridymite	131.0	1.379**	135.6	5.3
cristobalite	125.7	1.323	130.3	-

**But! The region that causes the bad fit is right where we had to do the most egregious stitching and scaling of the old lab data....

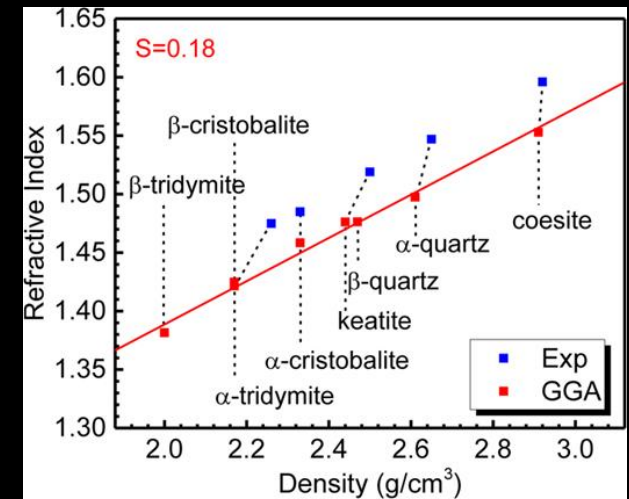
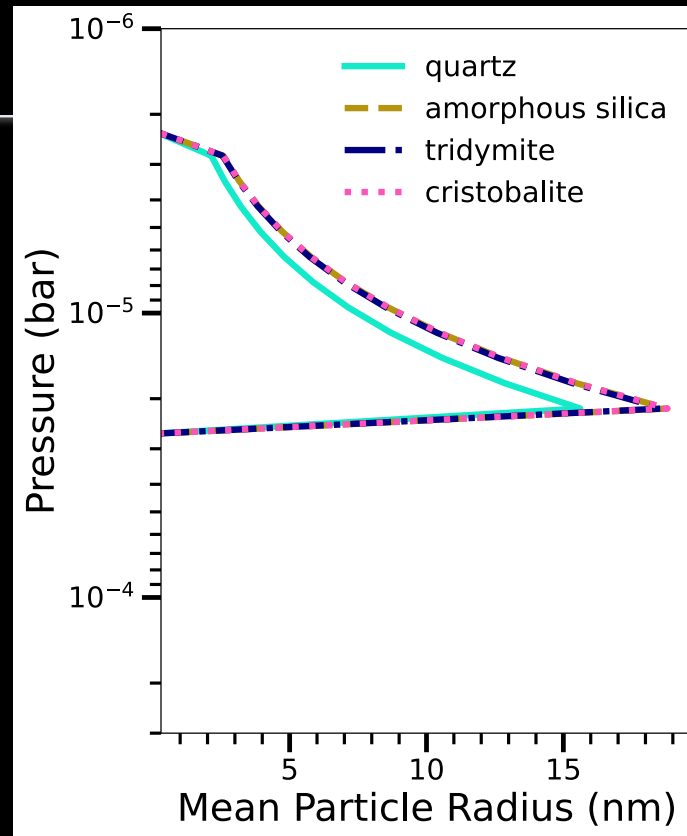
Density differences in silica polymorph particles cause further divergences but **very minorly**



Model	χ^2	χ^2/n	BIC	Δ BIC
<i>Mie coefficients, BIC k = 1, n = 95, dof = 94</i>				
quartz	126.3	1.330	130.9	0.6
amorphous silica	126.0	1.327	130.6	0.3
tridymite	131.0	1.379**	135.6	5.3
cristobalite	125.7	1.323	130.3	-
<i>Mie coefficients and density, BIC k = 2, n = 95, dof = 93</i>				
quartz	126.3	1.330	135.4	5.1
amorphous silica	125.4	1.320	134.5	4.2
tridymite	131.9	1.389	141.0	10.1
cristobalite	125.2	1.318	134.3	4.0

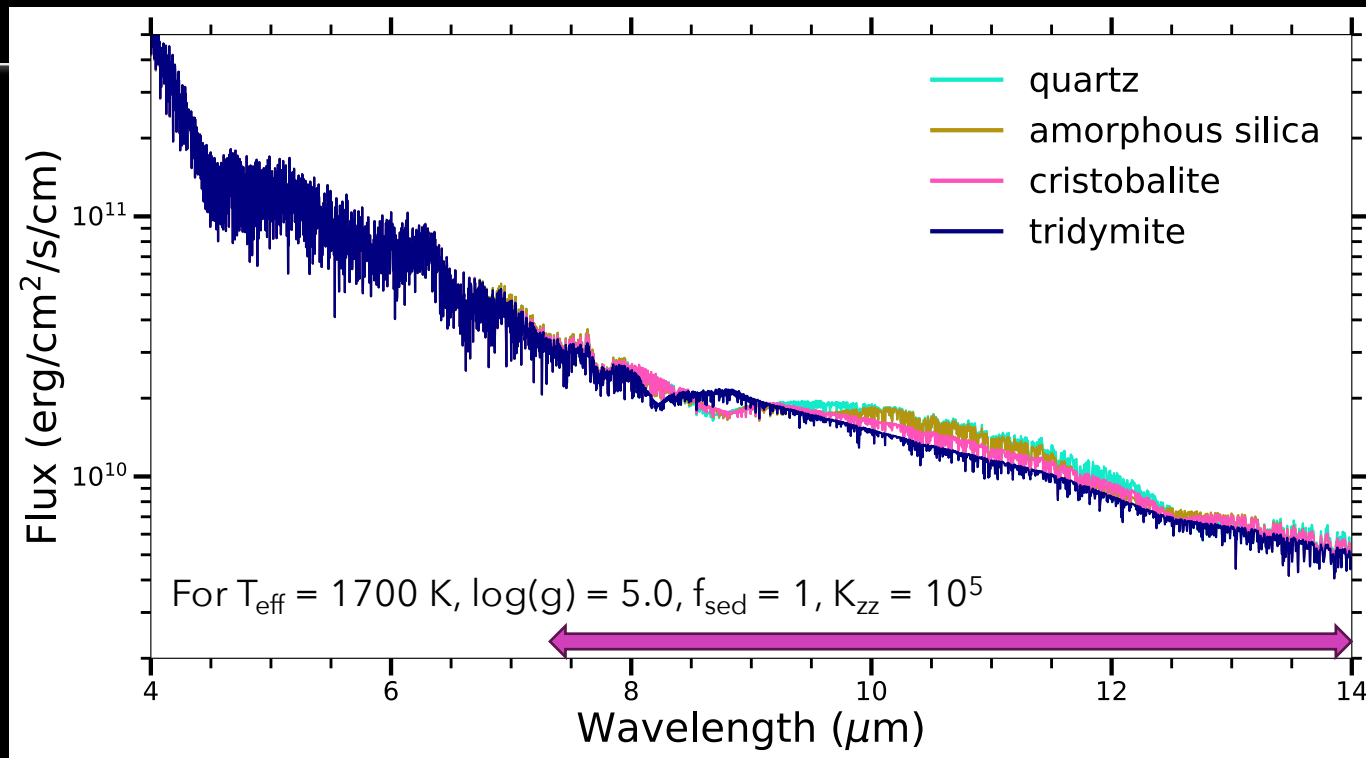
Moran, Marley, Crossley, *ApJL*, 2024

Polymorph densities aren't different enough to drastically change the particle size distributions



Chen et al., *J. Appl. Phys.*, 2023

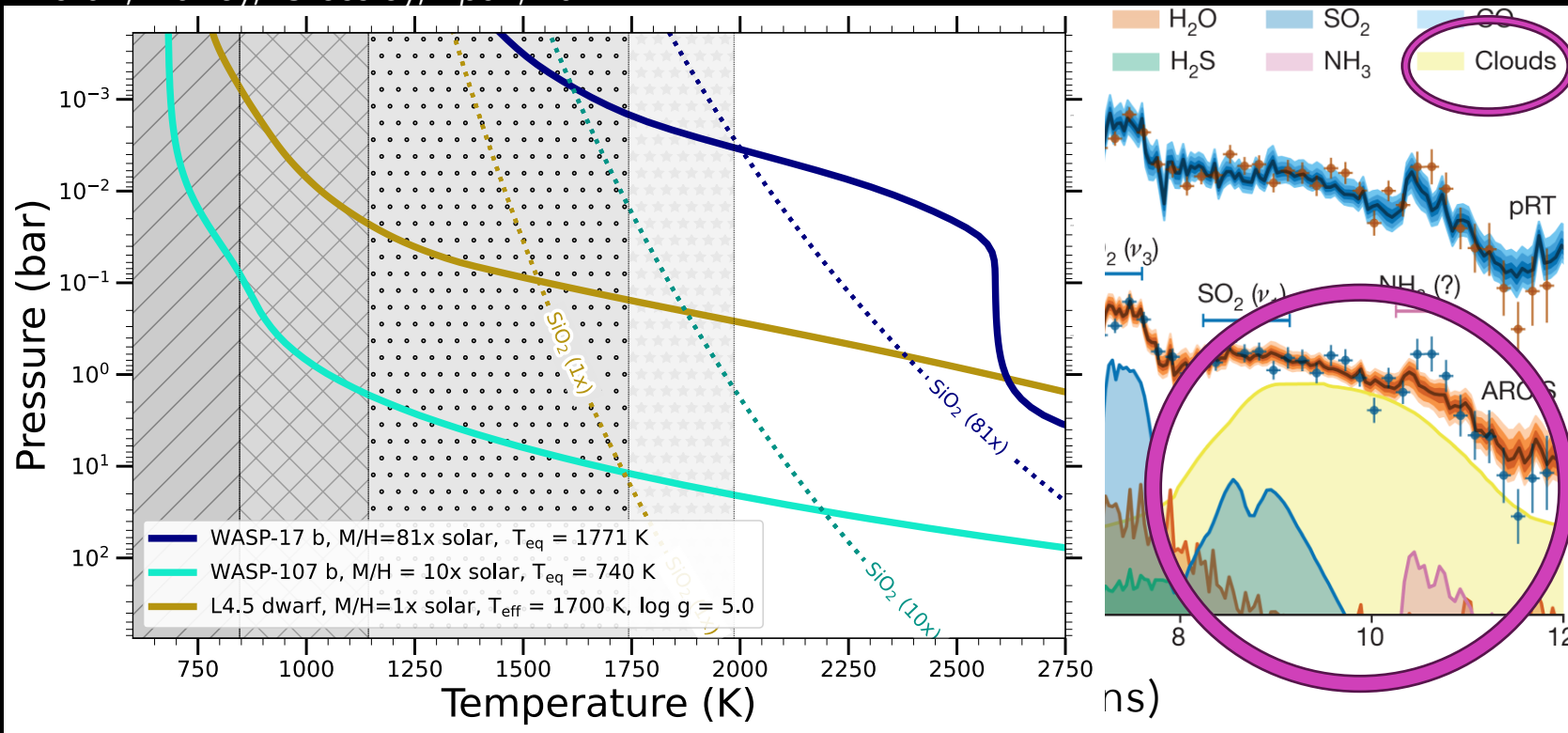
For a “typical” L-dwarf, cristobalite and tridymite particles affect emission spectra at JWST wavelengths if particle size distributions are narrow



Moran, Marley, & Crossley, *ApJL*, 2024

Case study: WASP-107 b and liquid, crystal, or amorphous cloud particles

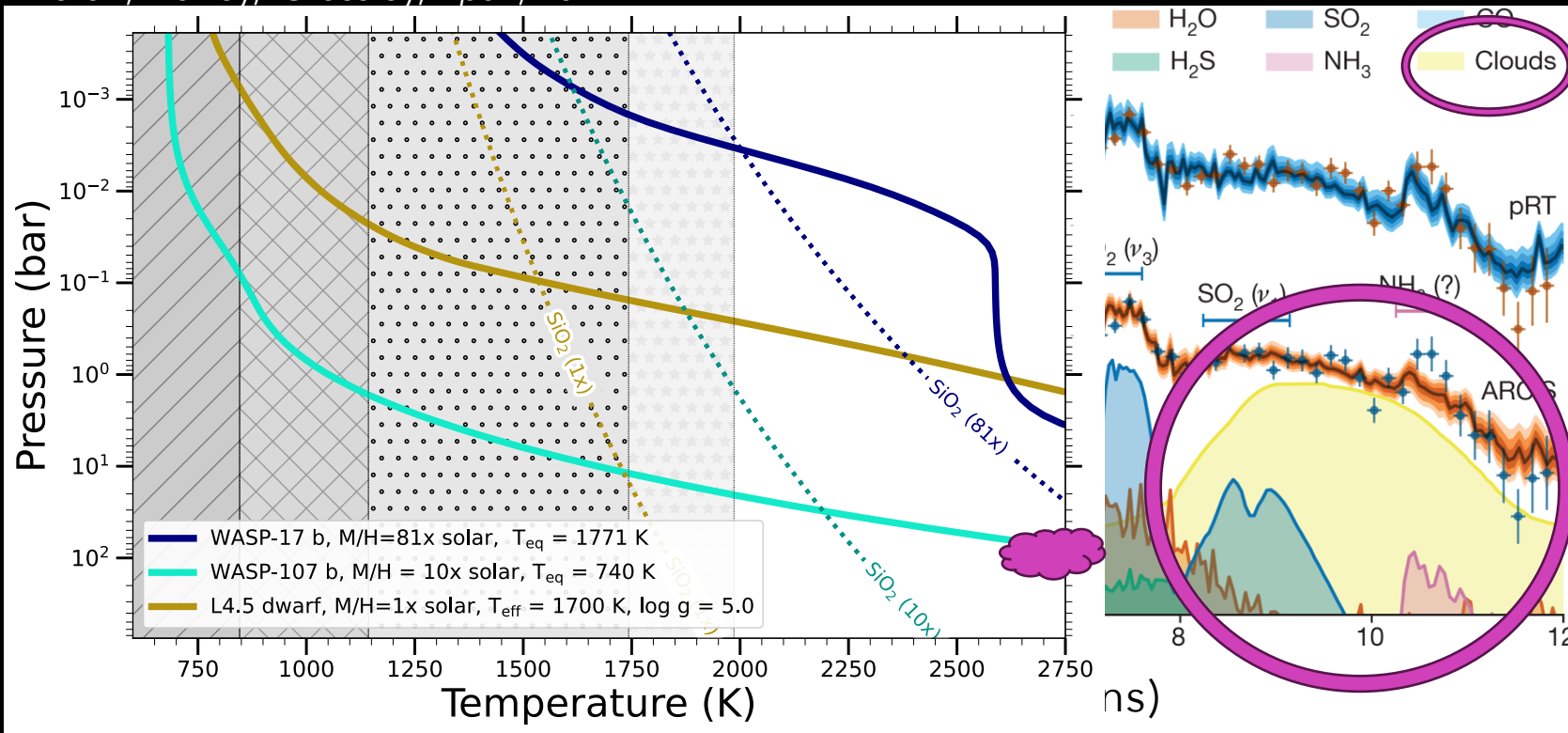
Moran, Marley, Crossley, *ApJL*, 2024



Dyrek et al., *Nature*, 2024

Case study: WASP-107 b and liquid, crystal, or amorphous cloud particles

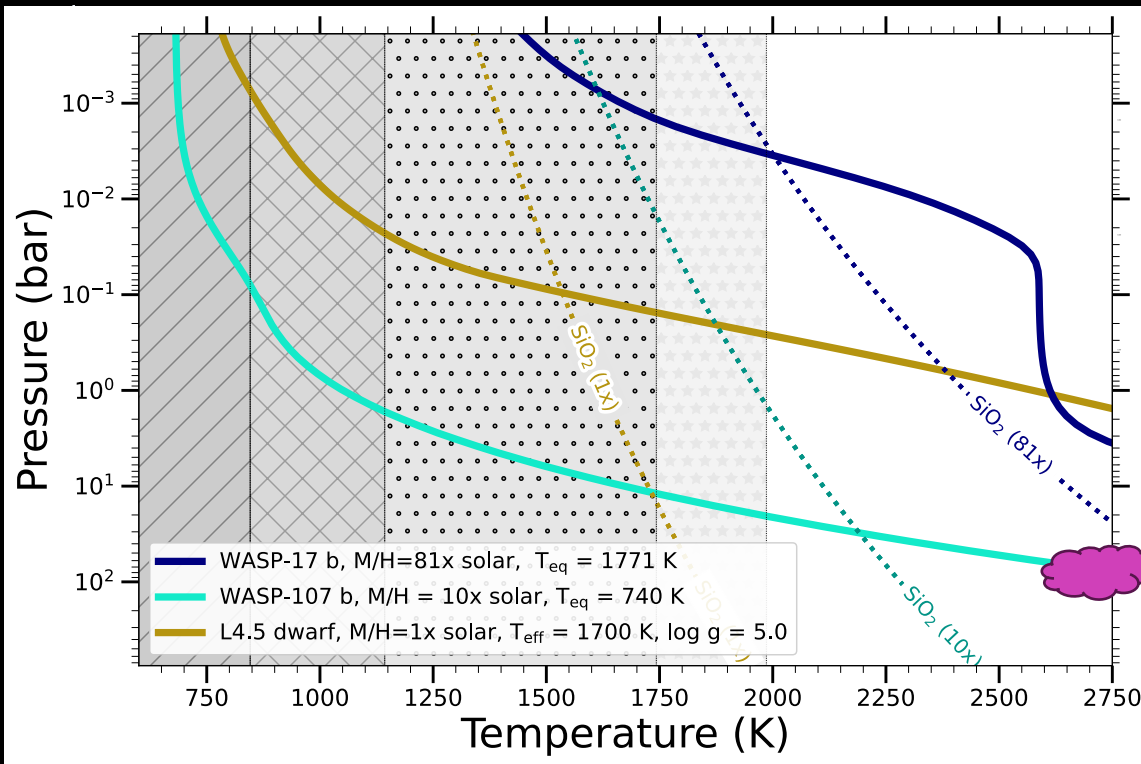
Moran, Marley, Crossley, *ApJL*, 2024



Dyrek et al., *Nature*, 2024

Case study: WASP-107 b and liquid, crystal, or amorphous cloud particles

Moran, Marley, Crossley, *ApJL*, 2024



$$\tau_{dyn} \approx \frac{H}{K_{zz}}$$

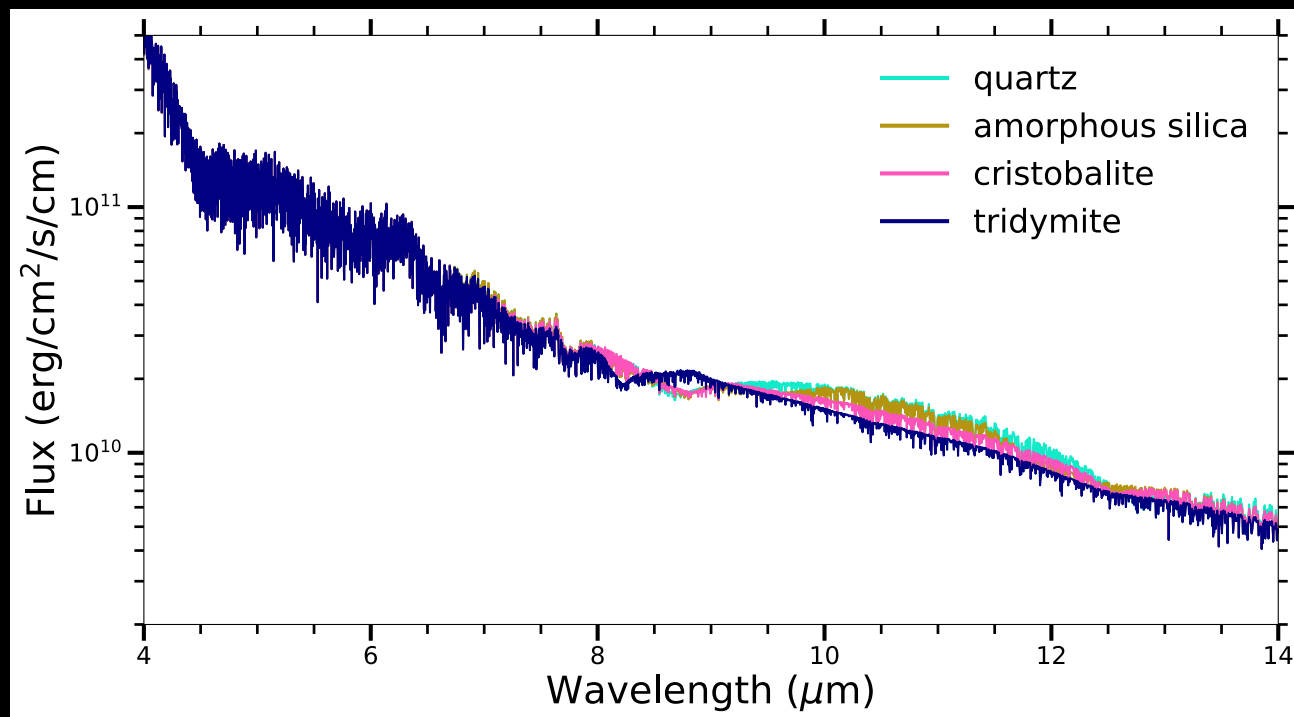
$$H = 800 \text{ km}, K_{zz} = 10^{12} \text{ cm/s}^2$$

$$\tau_{dyn} = 1.3 \times 10^4 \text{ s} = 3.5 \text{ hours}$$

→ particle speed = 220 km/h
(Cat 4 hurricane, straight up!)

30 bars to 1 mbar = 10 scale heights
→ 40ish hours for our cloud to rise!

Mixed crystal phase clouds are thus likely for a variety of atmospheres...
potentially explaining amorphous signatures?



An abbreviated list of things I think this impacts...

1. Lab work needed to refine all this!
 2. Lava worlds...
 3. Other mineral polymorphs...!
 4. Global circulation models and atmospheric dynamics...
-

Takeaways

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SiO₂ discoveries and JWST have ushered in a **new era** of (silicate) cloud studies

Accounting for **polymorphs** takes clouds a tracers of physics and chemistry to a new level

One of the more stable phases of SiO₂ at WASP-17b conditions better matches **JWST data**

There's a lot of exciting **experimental, theoretical, and observational follow-up** here

→ We are proposing a silicate condensation sequence based not just in chemistry, but also in **mineralogy**