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#### @methanoJen www.JenniferGlass.com

2019 Sagan Exoplanet Summer Workshop "Astrobiology for Astronomers" July 15, 2019

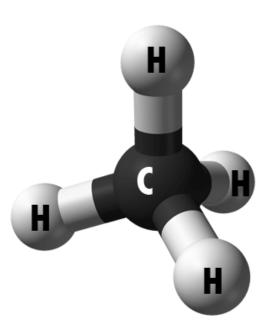
> Artwork by Sam Walton

# On Earth, O<sub>2</sub> and CH<sub>4</sub> have no significant abiotic sources

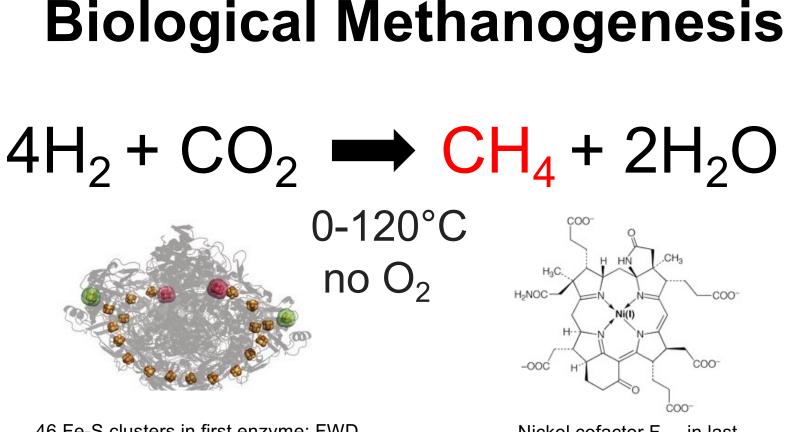


N<sub>2</sub> 78.08%
O<sub>2</sub> 20.95%
Ar 0.93%
CO<sub>2</sub> 0.0407%
Ne 0.0018%
He 0.000524%
CH<sub>4</sub> 0.00018%

# I. Modern







46 Fe-S clusters in first enzyme: FWD Wagner et al., 2016, *Science* 

Nickel cofactor F<sub>430</sub> in last enzyme (MCR)

# Abiotic Methane Production $4H_2 + CO_2 \longrightarrow CH_4 + 2H_2O$ "Fischer-Tropsch Process" $3H_2 + CO \longrightarrow CH_4 + H_2O$



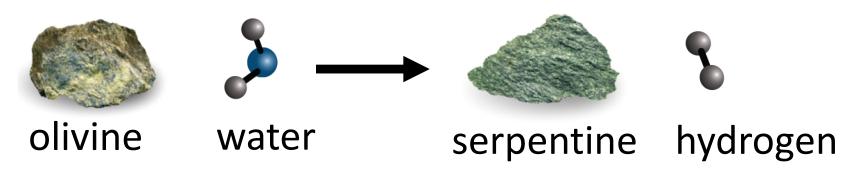
150–300 °C

Abiotic reactions are sluggish at low temperatures without mineral catalysts (Seewald et al. 2006, McCollom 2016)

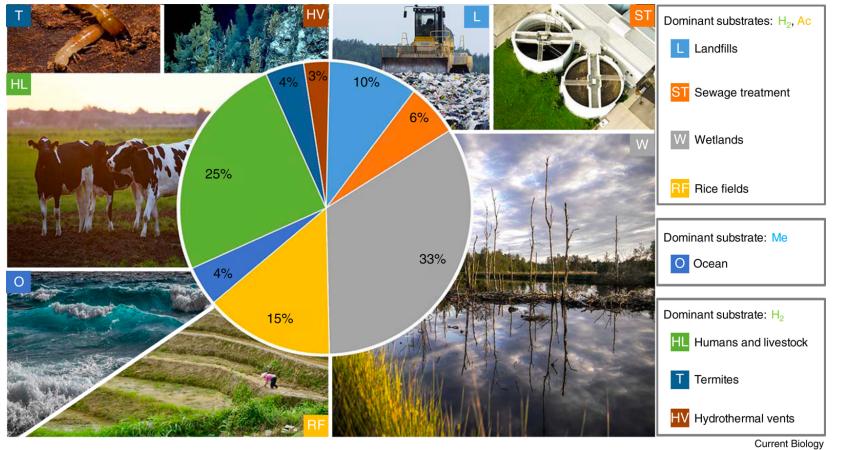
# Methane production is overwhelmingly biological

"Abiotic methane production estimates from serpentinization ranging between approximately 1/30th and 1/150th the present biotic flux appear reasonable for modern Earth." - Arney et al., 2018, <u>Astrobiology</u>

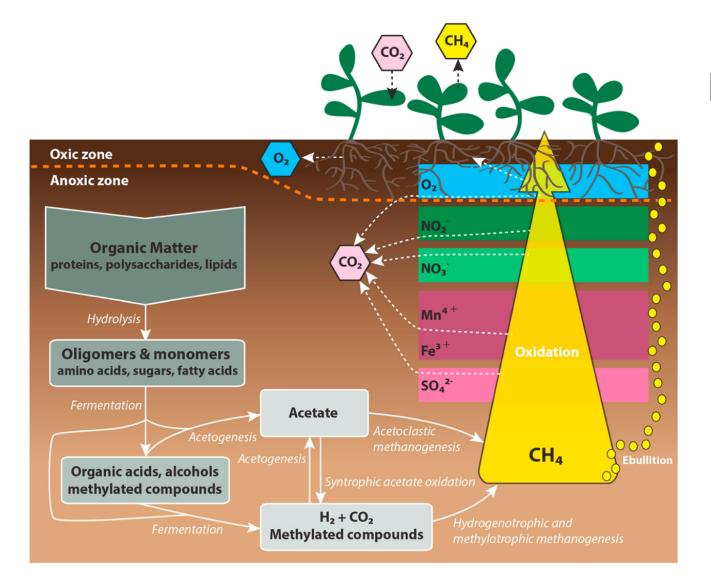
## Serpentinization: Source of Hydrogen



#### **Biological methane emissions to the atmosphere**



Lyu Z, Shao N, Akinyemi T, Whitman WB (2018) Methanogenesis. Current Biology 28: R727-R732



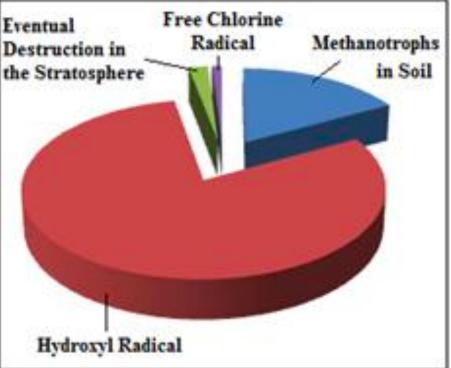
#### Methane cycling prior to release to atmosphere

Dean JF, Middelburg JJ, Röckmann T., Aerts R, Blauw LG, Egger M, et al (2018). Methane feedbacks to the global climate system in a warmer world. <u>Reviews of</u> <u>Geophysics</u>, 56, 207–250.

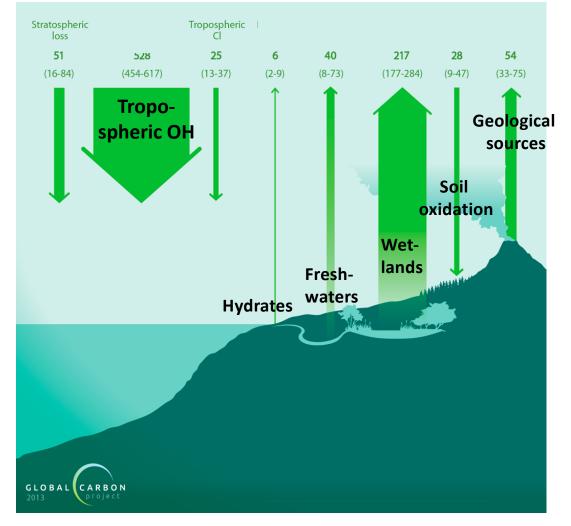
# The major sinks for atmospheric methane is photolytic destruction, which depends on OH radicals

Modern methane's atmospheric lifetime is ~10 years

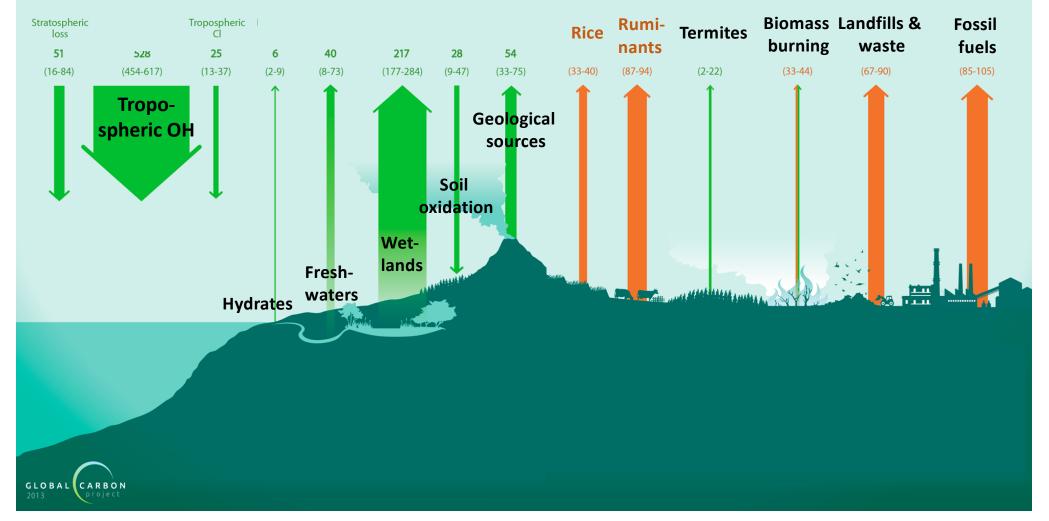
May not go quite as fast in atmosphere without oxygen



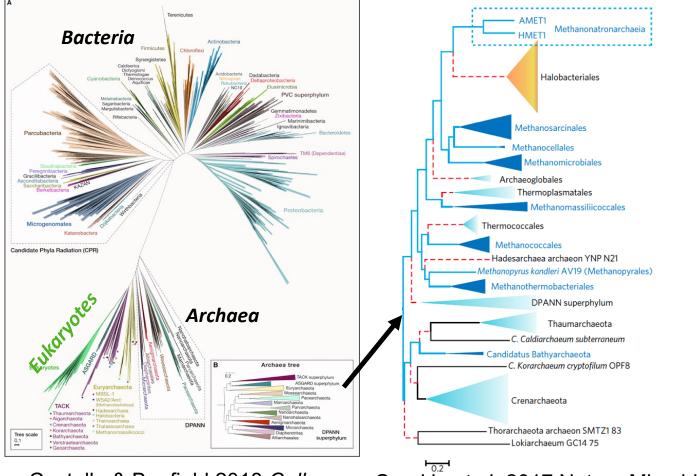
#### **Natural Methane Sources and Sinks (Tg yr<sup>-1</sup>)**



#### Anthropogenic Methane Sources and Sinks (Tg yr<sup>-1</sup>)



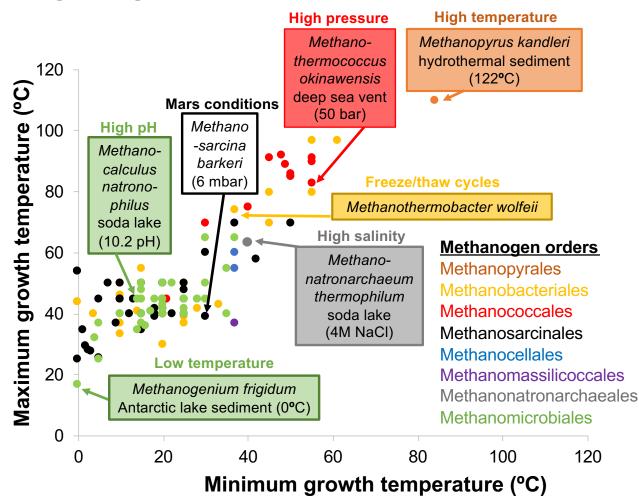
# **Microbial methane makers: Euryarchaeota**



Castelle & Banfield 2018 Cell

Sorokin et al. 2017 Nature Microbiol

#### Methanogens grow at extreme environmental conditions



Glass & Whitman, 2019, in press, Methanogenesis, Encyclopedia of Astrobiology



#### ARTICLE

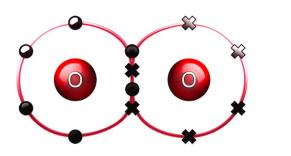
DOI: 10.1038/s41467-018-02876-y OPEN

#### Biological methane production under putative Enceladus-like conditions

Ruth-Sophie Taubner<sup>1,2</sup>, Patricia Pappenreiter<sup>3</sup>, Jennifer Zwicker<sup>4</sup>, Daniel Smrzka<sup>4</sup>, Christian Pruckner<sup>1</sup>, Philipp Kolar<sup>1</sup>, Sébastien Bernacchi<sup>5</sup>, Arne H. Seifert<sup>5</sup>, Alexander Krajete<sup>5</sup>, Wolfgang Bach<sup>6</sup>, Jörn Peckmann <sup>4,7</sup>, Christian Paulik <sup>3</sup>, Maria G. Firneis<sup>2</sup>, Christa Schleper<sup>1</sup> & Simon K.-M.R. Rittmann <sup>1</sup>

The detection of silica-rich dust particles, as an indication for ongoing hydrothermal activity, and the presence of water and organic molecules in the plume of Enceladus, have made Saturn's icy moon a hot spot in the search for potential extraterrestrial life. Methanogenic archaea are among the organisms that could potentially thrive under the predicted conditions on Enceladus, considering that both molecular hydrogen (H<sub>2</sub>) and methane (CH<sub>4</sub>) have been detected in the plume. Here we show that a methanogenic archaeon, *Methanothermococcus okinawensis*, can produce CH<sub>4</sub> under physicochemical conditions extrapolated for Enceladus. Up to 72% carbon dioxide to CH<sub>4</sub> conversion is reached at 50 bar in the presence of potential inhibitors. Furthermore, kinetic and thermodynamic computations of low-temperature serpentinization indicate that there may be sufficient H<sub>2</sub> gas production to serve as a substrate for CH<sub>4</sub> production on Enceladus. We conclude that some of the CH<sub>4</sub> detected in the plume of Enceladus might, in principle, be produced by methanogens.

# I. Modern





# Take another deep breath

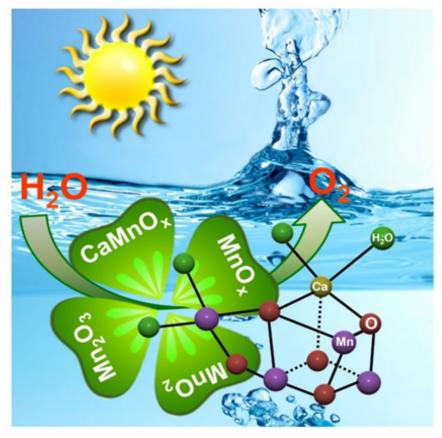


... and thank



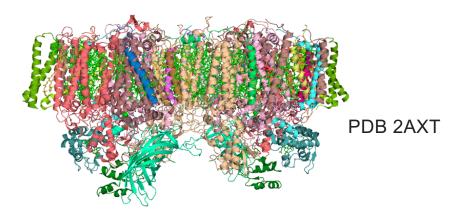
for the O<sub>2</sub> you're breathing

#### Photosystem II: nature's O<sub>2</sub> producer



Manganese-rich oxygen-evolving complex

Originated in cyanobacteria at least 2.4 billion years ago, spread to algae and plants by endosymbiosis



#### Large biological O<sub>2</sub> fluxes are balanced

#### Photosynthesis:

Process by which  $CO_2$  is converted to organic carbon (simplified as  $CH_2O$ ) using energy from sunlight:

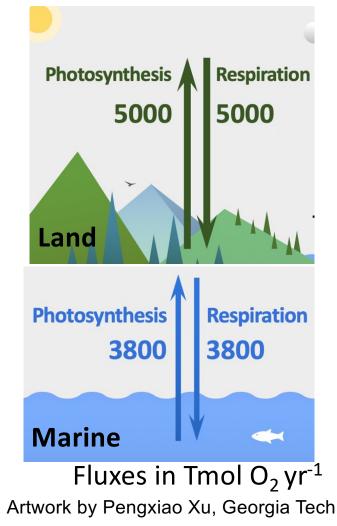
 $CO_2 + H_2O + sunlight + nutrients \rightarrow CH_2O + O_2$ 

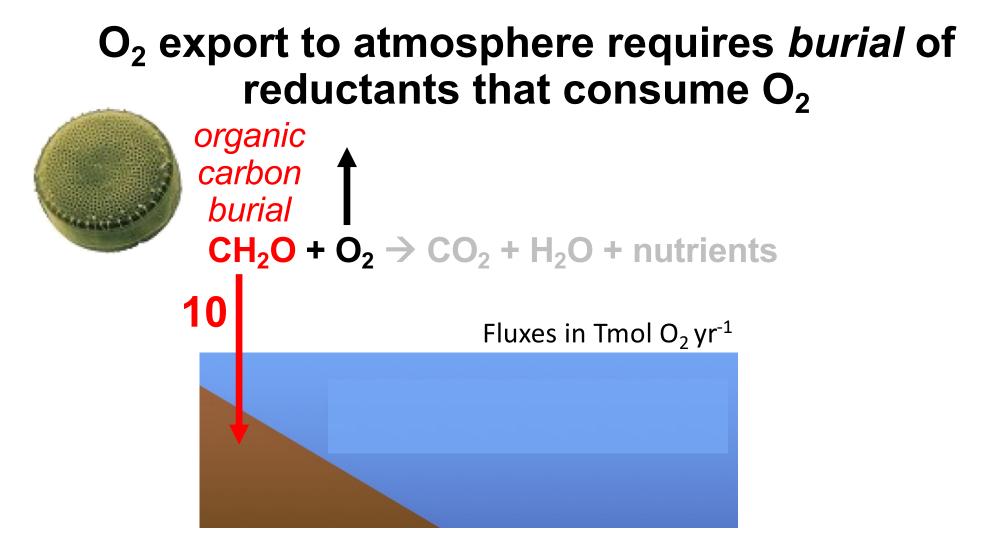
# Aerobic Respiration:

Process by which organic carbon is oxidized to  $CO_2$  with  $O_2$  to fuel ATP production:

#### $CH_2O + O_2 \rightarrow CO_2 + H_2O + nutrients$

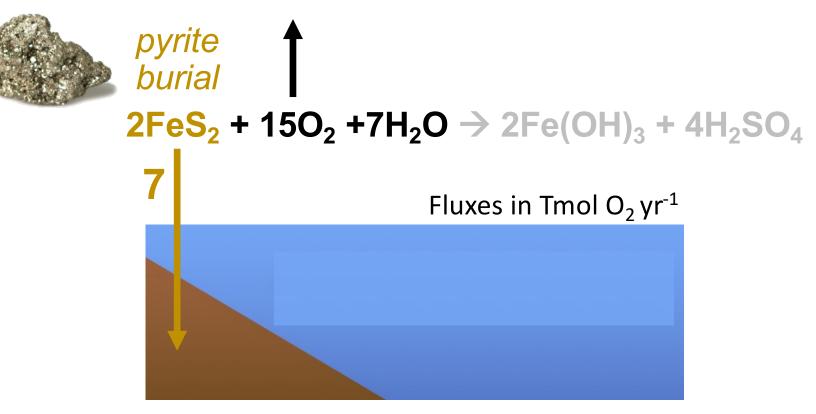
Kasting, JF, Canfield DE (2012) The Global Oxygen Cycle. Chapter 7. <u>Fundamentals of Geobiology</u>, 93-104.



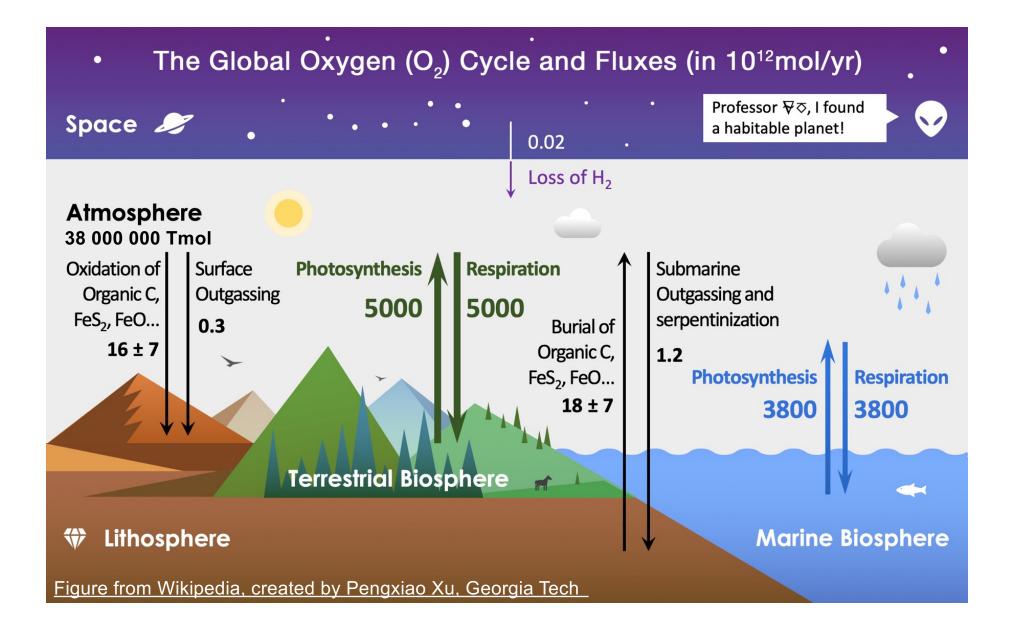


Kasting, JF, Canfield DE (2012) The Global Oxygen Cycle. Chapter 7. Fundamentals of Geobiology, 93-104.

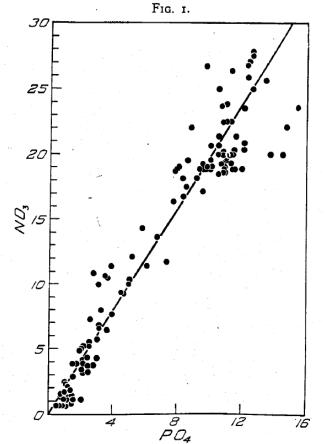
# O<sub>2</sub> export to atmosphere requires *burial* of reductants that consume O<sub>2</sub>



Kasting, JF, Canfield DE (2012) The Global Oxygen Cycle. Chapter 7. Fundamentals of Geobiology, 93-104.



# "Redfield Ratio"



Consistent atomic ratio of **106 C: 16 N: 1P** in marine phytoplankton.

Leaves its imprint on ocean chemistry

Now extended to include terrestrial life and trace elements

Redfield, A.C. (1934) **On the Proportions of Organic Derivatives in Sea Water and Their Relation to the Composition of Plankton**. James Johnstone Memorial Volume, University Press of Liverpool, 176-192.

Correlation between concentrations of nitrate and phosphate in the waters of western Atlantic Ocean. Ordinate, concentration of nitrate, units  $10^{-3}$  millimols per liter; abscissa, concentration of phosphate, units  $10^{-4}$  millimols per liter. The line represents a ratio of  $\triangle N : \triangle P = 20$ : I milligram atoms.

# **Nutrient limitation**

#### Baking a cake

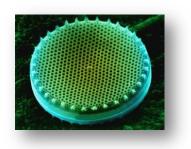
4 eggs:
3 cups flour:
2 cups sugar
+ (trace butter,
baking powder,
vanilla extract,
salt, etc.)



*If you only have 4 eggs, even if you have infinite flour and sugar, you can only make 1 cake.* 

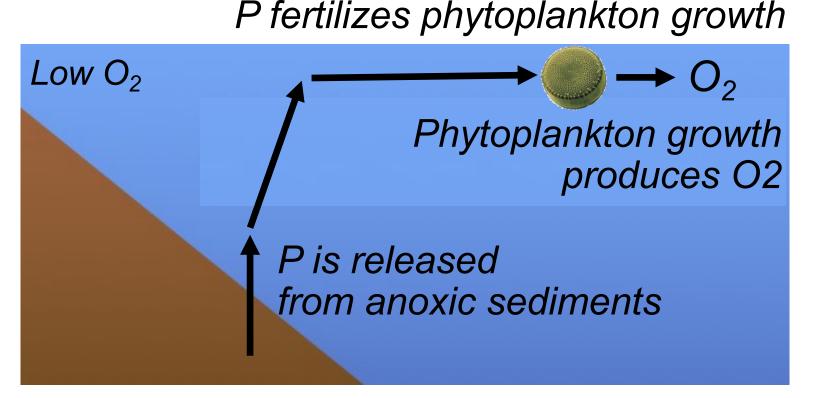
#### Making an organism

106 moles carbon:
16 moles nitrogen:
1 moles phosphorus
+ (trace iron,
manganese,
molybdenum,
zinc, etc.)



If you only have 1 mol of P, even if you have infinite nitrogen and carbon, you can only make 1 diatom (equivalent).

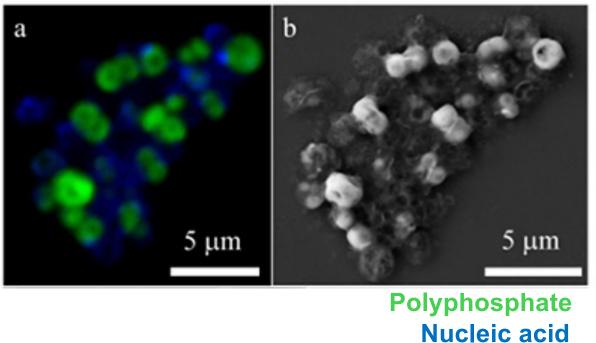
## Phosphorus availability may control long-term O<sub>2</sub>



Catling D, Zahnle Z (2003). Evolution of Atmospheric Oxygen. Encyclopedia of Atmospheric Sciences, pp. 754-76

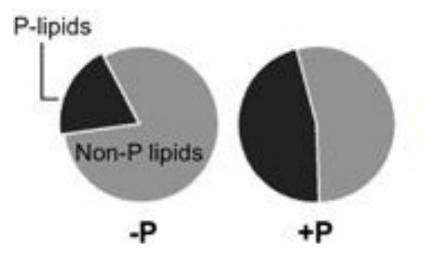
#### Life is extremely good at acquiring scarce bioessential elements

# I. Storage



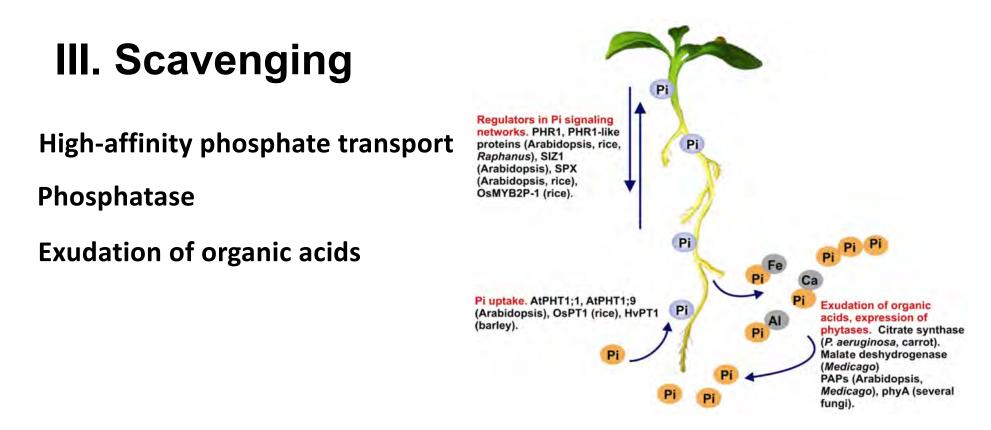
Rivas-Lamelo et al 2017 Magnetotactic bacteria as a new model for P sequestration in the ferruginous Lake Pavin. *Geochemical Perspectives Letters*  Life is extremely good at acquiring scarce bioessential elements

# **II.** Substitution

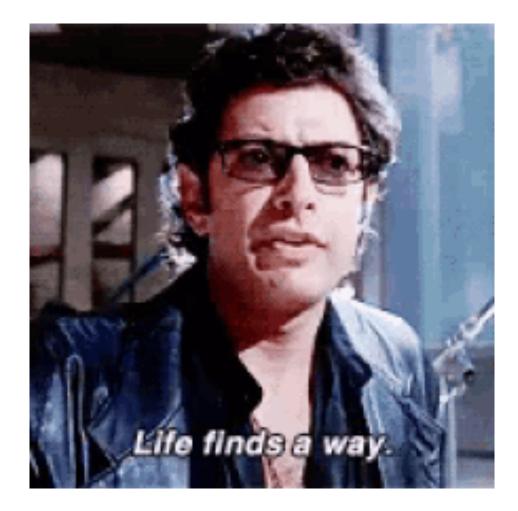


Sebastian et al. 2016, Lipid remodelling is a widespread strategy in marine heterotrophic bacteria upon phosphorus deficiency. ISME Journal

#### Life is extremely good at acquiring scarce bioessential elements



Lopez-Arredondo et al. 2013, Biotechnology of nutrient uptake and assimilation in plants. Int. J. of Developmental Biology



# **Recommended References**

#### Modern Methane Cycle

Lyu Z, Shao N, Akinyemi T, Whitman WB (2018) Methanogenesis. <u>Current Biology 2</u>8: R727-R732

Dean JF, Middelburg JJ, Röckmann T., Aerts R, Blauw LG, Egger M, et al (2018). Methane feedbacks to the global climate system in a warmer world. <u>Reviews of Geophysics</u>, 56, 207–250.

#### Modern Oxygen Cycle

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Catling D, Zahnle Z (2003). Evolution of Atmospheric Oxygen. <u>Encyclopedia of Atmospheric Sciences</u>, pp. 754-76.

#### **Biosignature Gases**

Seager S, M Schrenk, W Brazelton. An Astrophysical View of Earth-Based Biosignature Gases. Astrobiology 12: 61-82